

# Benzene, 1-(1,1-dimethylethyl)-2,4-dimethyl

Inchi:	InChI=1S/C12H18/c1-9-6-7-11(10(2)8-9)12(3,4)5/h6-8H,1-5H3
InchiKey:	UAGBDAUMTJRHBN-UHFFFAOYSA-N
Formula:	C12H18
SMILES:	Cc1ccc(C(C)(C)C)c(C)c1
Mol. weight [g/mol]:	162.27

## Physical Properties

Property code	Value	Unit	Source
gf	146.15	kJ/mol	Joback Method
hf	-86.17	kJ/mol	Joback Method
hfus	12.69	kJ/mol	Joback Method
hvap	44.61	kJ/mol	Joback Method
log10ws	-3.74		Crippen Method
logp	3.601		Crippen Method
mcvol	156.180	ml/mol	McGowan Method
pc	2379.54	kPa	Joback Method
rinsol	1168.00		NIST Webbook
tb	507.37	K	Joback Method
tc	722.03	K	Joback Method
tf	278.88	K	Joback Method
vc	0.589	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	349.74	J/molxK	507.37	Joback Method
cpg	367.17	J/molxK	543.15	Joback Method
cpg	383.55	J/molxK	578.92	Joback Method
cpg	398.95	J/molxK	614.70	Joback Method
cpg	413.40	J/molxK	650.48	Joback Method
cpg	426.96	J/molxK	686.25	Joback Method
cpg	439.68	J/molxK	722.03	Joback Method
dvisc	0.0025008	Paxs	278.88	Joback Method
dvisc	0.0012488	Paxs	316.96	Joback Method

dvisc	0.0007238	Paxs	355.04	Joback Method
dvisc	0.0004663	Paxs	393.12	Joback Method
dvisc	0.0003246	Paxs	431.21	Joback Method
dvisc	0.0002397	Paxs	469.29	Joback Method
dvisc	0.0001852	Paxs	507.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R52993&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R52993&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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