

# Benzene, 1-ethyl-4-(1-methylpropyl)

Other names:	1-ethyl-4-sec-butylbenzene
Inchi:	InChI=1S/C12H18/c1-4-10(3)12-8-6-11(5-2)7-9-12/h6-10H,4-5H2,1-3H3
InchiKey:	WCBFCOJVLMFJHZ-UHFFFAOYSA-N
Formula:	C12H18
SMILES:	CCc1ccc(C(C)CC)cc1
Mol. weight [g/mol]:	162.27

## Physical Properties

Property code	Value	Unit	Source
gf	150.50	kJ/mol	Joback Method
hf	-71.23	kJ/mol	Joback Method
hfus	16.96	kJ/mol	Joback Method
hvap	44.86	kJ/mol	Joback Method
log10ws	-3.88		Crippen Method
logp	3.763		Crippen Method
mcvol	156.180	ml/mol	McGowan Method
pc	2388.85	kPa	Joback Method
tb	505.18	K	Joback Method
tc	710.08	K	Joback Method
tf	248.94	K	Joback Method
vc	0.594	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	346.40	J/molxK	505.18	Joback Method
cpg	363.34	J/molxK	539.33	Joback Method
cpg	379.40	J/molxK	573.48	Joback Method
cpg	394.59	J/molxK	607.63	Joback Method
cpg	408.95	J/molxK	641.78	Joback Method
cpg	422.51	J/molxK	675.93	Joback Method
cpg	435.30	J/molxK	710.08	Joback Method
dvisc	0.0037035	Paxs	248.94	Joback Method
dvisc	0.0015687	Paxs	291.65	Joback Method

dvisc	0.0008275	Paxs	334.35	Joback Method
dvisc	0.0005046	Paxs	377.06	Joback Method
dvisc	0.0003403	Paxs	419.77	Joback Method
dvisc	0.0002468	Paxs	462.47	Joback Method
dvisc	0.0001890	Paxs	505.18	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R52854&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R52854&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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