

# 2,5-Dimethylphenyl methyl carbinol

<b>Other names:</b>	«alpha»,2,5-Trimethylbenzyl alcohol Benzenemethanol, «alpha»,2,5-trimethyl- 1-(2,5-Dimethylphenyl)ethanol benzyl alcohol, «alpha»,2,5-trimethyl-
<b>Inchi:</b>	InChI=1S/C10H14O/c1-7-4-5-8(2)10(6-7)9(3)11/h4-6,9,11H,1-3H3
<b>InchiKey:</b>	VHLZFCOCNJEXTQ-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	<chem>Cc1ccc(C)c(C(C)O)c1</chem>
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	32917-52-5

## Physical Properties

Property code	Value	Unit	Source
gf	-12.79	kJ/mol	Joback Method
hf	-193.65	kJ/mol	Joback Method
hfus	15.48	kJ/mol	Joback Method
hvap	57.75	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	2.357		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3166.83	kPa	Joback Method
tb	556.58	K	Joback Method
tc	755.02	K	Joback Method
tf	299.74	K	Joback Method
vc	0.500	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.37	J/molxK	556.58	Joback Method
cpg	323.63	J/molxK	589.65	Joback Method
cpg	335.25	J/molxK	622.73	Joback Method
cpg	346.27	J/molxK	655.80	Joback Method
cpg	356.70	J/molxK	688.88	Joback Method

cpg	366.57	J/mol×K	721.95	Joback Method
cpg	375.88	J/mol×K	755.02	Joback Method
dvisc	0.0084193	Paxs	299.74	Joback Method
dvisc	0.0024711	Paxs	342.55	Joback Method
dvisc	0.0009523	Paxs	385.35	Joback Method
dvisc	0.0004441	Paxs	428.16	Joback Method
dvisc	0.0002379	Paxs	470.97	Joback Method
dvisc	0.0001414	Paxs	513.77	Joback Method
dvisc	0.0000911	Paxs	556.58	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C32917525&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C32917525&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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