

(.+/-)-Lavandulol, chlorodifluoroacetate

Inchi:	InChI=1S/C12H17ClF2O2/c1-8(2)5-6-10(9(3)4)7-17-11(16)12(13,14)15/h5,10H,3,6-7H2,
InchiKey:	AYASINNVPARBCO-UHFFFAOYSA-N
Formula:	C12H17ClF2O2
SMILES:	<chem>C=C(C)C(CC=C(C)C)COC(=O)C(F)(F)Cl</chem>
Mol. weight [g/mol]:	266.71

Physical Properties

Property code	Value	Unit	Source
gf	-433.95	kJ/mol	Joback Method
hf	-734.73	kJ/mol	Joback Method
hfus	25.34	kJ/mol	Joback Method
hvap	51.98	kJ/mol	Joback Method
log10ws	-4.14		Crippen Method
logp	3.910		Crippen Method
mvol	194.560	ml/mol	McGowan Method
pc	1857.91	kPa	Joback Method
tb	583.15	K	Joback Method
tc	770.17	K	Joback Method
tf	280.92	K	Joback Method
vc	0.762	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	475.39	J/molxK	583.15	Joback Method
cpg	489.58	J/molxK	614.32	Joback Method
cpg	502.95	J/molxK	645.49	Joback Method
cpg	515.55	J/molxK	676.66	Joback Method
cpg	527.42	J/molxK	707.83	Joback Method
cpg	538.60	J/molxK	739.00	Joback Method
cpg	549.12	J/molxK	770.17	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U376229&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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