

# 1,9-Decadiyne

<b>Other names:</b>	deca-1,9-diyne
<b>Inchi:</b>	InChI=1S/C10H14/c1-3-5-7-9-10-8-6-4-2/h1-2H,5-10H2
<b>InchiKey:</b>	ILVDYAGPHFWNQI-UHFFFAOYSA-N
<b>Formula:</b>	C10H14
<b>SMILES:</b>	C#CCCCCCCC#C
<b>Mol. weight [g/mol]:</b>	134.22
<b>CAS:</b>	1720-38-3

## Physical Properties

Property code	Value	Unit	Source
gf	479.46	kJ/mol	Joback Method
hf	334.07	kJ/mol	Joback Method
hfus	27.61	kJ/mol	Joback Method
hvap	37.57	kJ/mol	Joback Method
log10ws	-3.60		Crippen Method
logp	2.593		Crippen Method
mcvol	134.560	ml/mol	McGowan Method
pc	2823.32	kPa	Joback Method
tb	408.44	K	Joback Method
tc	594.16	K	Joback Method
tf	296.40	K	Joback Method
vc	0.519	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.65	J/molxK	408.44	Joback Method
cpg	271.33	J/molxK	439.39	Joback Method
cpg	283.40	J/molxK	470.35	Joback Method
cpg	294.88	J/molxK	501.30	Joback Method
cpg	305.80	J/molxK	532.25	Joback Method
cpg	316.18	J/molxK	563.21	Joback Method
cpg	326.04	J/molxK	594.16	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	352.00 ± 1.00	K	1.70	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.49481e+01
Coeff. B	-4.15561e+03
Coeff. C	-6.82230e+01
Temperature range (K), min.	351.68
Temperature range (K), max.	499.45

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1720383&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1720383&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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