2,2,4-trimethylcyclohexene-carboxaldehyde

Inchi: InChI=1S/C10H16O/c1-8-4-5-9(7-11)10(2,3)6-8/h5,7-8H,4,6H2,1-3H3
InchiKey: PWFBZZTXVTYUMA-UHFFFAOYSA-N
Formula: C10H16O
SMILES: CC1CC=C(C=O)C(C)(C)C1
Mol. weight [g/mol]: 152.23

Physical Properties

Property code	Value	Unit	Source
gf	-34.62	kJ/mol	Joback Method
hf	-239.78	kJ/mol	Joback Method
hfus	11.39	kJ/mol	Joback Method
hvap	44.50	kJ/mol	Joback Method
log10ws	-2.55		Crippen Method
logp	2.568		Crippen Method
mcvol	138.170	ml/mol	McGowan Method
pc	2871.95	kPa	Joback Method
ripol	1715.00		NIST Webbook
tb	496.12	K	Joback Method
tc	709.88	K	Joback Method
tf	284.78	K	Joback Method
vc	0.528	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.56	J/mol×K	496.12	Joback Method
cpg	330.31	J/mol×K	531.75	Joback Method
cpg	346.01	J/mol×K	567.37	Joback Method
cpg	360.76	J/mol×K	603.00	Joback Method
cpg	374.68	J/mol×K	638.62	Joback Method
cpg	387.85	J/mol×K	674.25	Joback Method
cpg	400.37	J/mol×K	709.88	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R336484&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume