

# Linden ether

<b>Inchi:</b>	InChI=1S/C10H14O/c1-7-3-4-9-8(2)6-11-10(9)5-7/h5,10H,3-4,6H2,1-2H3
<b>InchiKey:</b>	NDGVBBGSPLJJRM-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	CC1=CC2OCC(C)=C2CC1
<b>Mol. weight [g/mol]:</b>	150.22

## Physical Properties

Property code	Value	Unit	Source
gf	71.14	kJ/mol	Joback Method
hf	-153.12	kJ/mol	Joback Method
hfus	19.81	kJ/mol	Joback Method
hvap	45.59	kJ/mol	Joback Method
log10ws	-2.70		Crippen Method
logp	2.442		Crippen Method
mvol	127.310	ml/mol	McGowan Method
pc	3114.04	kPa	Joback Method
rinpol	1247.00		NIST Webbook
rinpol	1247.00		NIST Webbook
tb	499.37	K	Joback Method
tc	720.11	K	Joback Method
tf	297.67	K	Joback Method
vc	0.479	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	289.58	J/molxK	499.37	Joback Method
cpg	305.68	J/molxK	536.16	Joback Method
cpg	320.79	J/molxK	572.95	Joback Method
cpg	334.98	J/molxK	609.74	Joback Method
cpg	348.28	J/molxK	646.53	Joback Method
cpg	360.75	J/molxK	683.32	Joback Method
cpg	372.44	J/molxK	720.11	Joback Method
dvisc	0.0017208	Paxs	297.67	Joback Method

dvisc	0.0012280	Paxs	331.29	Joback Method
dvisc	0.0009326	Paxs	364.90	Joback Method
dvisc	0.0007418	Paxs	398.52	Joback Method
dvisc	0.0006115	Paxs	432.14	Joback Method
dvisc	0.0005183	Paxs	465.75	Joback Method
dvisc	0.0004492	Paxs	499.37	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R607315&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R607315&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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