

# Benzene, (1-methoxy-1-methylethyl)-

<b>Other names:</b>	Ether, «alpha», «alpha»-dimethylbenzyl methyl «alpha», «alpha»-Dimethylbenzyl methyl ether Methyl cumyl ether 2-Methoxy-2-phenylpropane (1-methoxy-1-methylethyl)benzene
<b>Inchi:</b>	InChI=1S/C10H14O/c1-10(2,11-3)9-7-5-4-6-8-9/h4-8H,1-3H3
<b>InchiKey:</b>	MRFQFQYRTNGOCZ-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O
<b>SMILES:</b>	COC(C)(C)c1ccccc1
<b>Mol. weight [g/mol]:</b>	150.22
<b>CAS:</b>	935-67-1

## Physical Properties

Property code	Value	Unit	Source
gf	43.57	kJ/mol	Joback Method
hf	-154.17	kJ/mol	Joback Method
hfus	9.47	kJ/mol	Joback Method
hvap	41.24	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	2.568		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	2937.70	kPa	Joback Method
tb	474.07	K	Joback Method
tc	690.87	K	Joback Method
tf	253.53	K	Joback Method
vc	0.494	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	283.83	J/mol×K	474.07	Joback Method
cpg	299.88	J/mol×K	510.20	Joback Method
cpg	314.94	J/mol×K	546.34	Joback Method
cpg	329.04	J/mol×K	582.47	Joback Method

cpg	342.23	J/mol×K	618.60	Joback Method
cpg	354.55	J/mol×K	654.73	Joback Method
cpg	366.04	J/mol×K	690.87	Joback Method
dvisc	0.0039926	Paxs	253.53	Joback Method
dvisc	0.0017530	Paxs	290.29	Joback Method
dvisc	0.0009261	Paxs	327.04	Joback Method
dvisc	0.0005566	Paxs	363.80	Joback Method
dvisc	0.0003673	Paxs	400.56	Joback Method
dvisc	0.0002599	Paxs	437.31	Joback Method
dvisc	0.0001940	Paxs	474.07	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C935671&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C935671&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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