

# Phenanthrene, 2-dodecyltetradecahydro-

<b>Other names:</b>	2-n-Dodecylperhydrophenanthrene 2-n-Dodecyl(tetradecahydrophenanthrene) 5-dodecyltricyclo[8.4.0.0(2,6)]tetracedane
<b>Inchi:</b>	InChI=1S/C26H48/c1-2-3-4-5-6-7-8-9-10-11-14-22-17-20-26-24(21-22)19-18-23-15-12-1
<b>InchiKey:</b>	YIZZECQVSWASEZ-UHFFFAOYSA-N
<b>Formula:</b>	C26H48
<b>SMILES:</b>	CCCCCCCCCCCC1CCC2C(CCC3CCCCC32)C1
<b>Mol. weight [g/mol]:</b>	360.66
<b>CAS:</b>	55334-22-0

## Physical Properties

Property code	Value	Unit	Source
gf	274.37	kJ/mol	Joback Method
hf	-433.05	kJ/mol	Joback Method
hfus	49.14	kJ/mol	Joback Method
hvap	73.45	kJ/mol	Joback Method
log10ws	-9.18		Crippen Method
logp	8.930		Crippen Method
mcvol	344.620	ml/mol	McGowan Method
pc	938.07	kPa	Joback Method
tb	826.51	K	Joback Method
tc	1026.96	K	Joback Method
tf	410.52	K	Joback Method
vc	1.321	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1203.39	J/molxK	826.51	Joback Method
cpg	1229.34	J/molxK	859.92	Joback Method
cpg	1253.73	J/molxK	893.33	Joback Method
cpg	1276.64	J/molxK	926.73	Joback Method
cpg	1298.15	J/molxK	960.14	Joback Method
cpg	1318.36	J/molxK	993.55	Joback Method

cpg	1337.35	J/mol×K	1026.96	Joback Method
dvisc	0.0030069	Paxs	410.52	Joback Method
dvisc	0.0016056	Paxs	479.85	Joback Method
dvisc	0.0010045	Paxs	549.18	Joback Method
dvisc	0.0006982	Paxs	618.51	Joback Method
dvisc	0.0005222	Paxs	687.85	Joback Method
dvisc	0.0004119	Paxs	757.18	Joback Method
dvisc	0.0003380	Paxs	826.51	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55334220&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55334220&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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