

# 4,18-dimethyl-tetracosane

<b>Inchi:</b>	InChI=1S/C26H54/c1-5-7-8-18-23-26(4)24-20-17-15-13-11-9-10-12-14-16-19-22-25(3)21
<b>InchiKey:</b>	KDVG MVXRRHIZTK-UHFFFAOYSA-N
<b>Formula:</b>	C26H54
<b>SMILES:</b>	CCCCCCC(C)CCCCCCCCCCCCCCC(C)CCC
<b>Mol. weight [g/mol]:</b>	366.71

## Physical Properties

Property code	Value	Unit	Source
gf	163.16	kJ/mol	Joback Method
hf	-590.53	kJ/mol	Joback Method
hfus	56.05	kJ/mol	Joback Method
hvap	72.69	kJ/mol	Joback Method
log10ws	-10.22		Crippen Method
logp	10.100		Crippen Method
mcvol	377.200	ml/mol	McGowan Method
pc	731.25	kPa	Joback Method
rinqol	2488.00		NIST Webbook
tb	793.40	K	Joback Method
tc	971.62	K	Joback Method
tf	352.78	K	Joback Method
vc	1.480	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1217.07	J/molxK	793.40	Joback Method
cpg	1326.13	J/molxK	941.92	Joback Method
cpg	1306.47	J/molxK	912.22	Joback Method
cpg	1285.78	J/molxK	882.51	Joback Method
cpg	1264.02	J/molxK	852.81	Joback Method
cpg	1241.13	J/molxK	823.10	Joback Method
cpg	1344.80	J/molxK	971.62	Joback Method
dvisc	0.0000369	Paxs	793.40	Joback Method
dvisc	0.0000531	Paxs	719.96	Joback Method

dvisc	0.0000831	Paxs	646.53	Joback Method
dvisc	0.0001457	Paxs	573.09	Joback Method
dvisc	0.0003014	Paxs	499.65	Joback Method
dvisc	0.0008010	Paxs	426.22	Joback Method
dvisc	0.0031975	Paxs	352.78	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R404676&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R404676&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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