

# dl-Camphoric anhydride

<b>Other names:</b>	3-Oxabicyclo[3.2.1]octane-2,4-dione, 1,8,8-trimethyl-, (.+/-)- DL-Camphoic anhydride (.+/-)-Camphoric anhydride
<b>Inchi:</b>	InChI=1S/C10H14O3/c1-5-4-6-8(11)13-9(12)7(5)10(6,2)3/h5-7H,4H2,1-3H3
<b>InchiKey:</b>	DLFUZOOSJJODPF-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O3
<b>SMILES:</b>	CC1CC2C(=O)OC(=O)C1C2(C)C
<b>Mol. weight [g/mol]:</b>	182.22
<b>CAS:</b>	595-30-2

## Physical Properties

Property code	Value	Unit	Source
gf	-221.59	kJ/mol	Joback Method
hf	-549.29	kJ/mol	Joback Method
hfus	16.57	kJ/mol	Joback Method
hvap	49.26	kJ/mol	Joback Method
log10ws	-1.47		Crippen Method
logp	1.368		Crippen Method
mcvol	139.050	ml/mol	McGowan Method
pc	2992.59	kPa	Joback Method
tb	603.71	K	Joback Method
tc	847.19	K	Joback Method
tf	495.00 ± 1.00	K	NIST Webbook
tt	493.90 ± 1.50	K	NIST Webbook
vc	0.524	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	389.50	J/molxK	603.71	Joback Method
cpg	407.72	J/molxK	644.29	Joback Method
cpg	424.95	J/molxK	684.87	Joback Method
cpg	441.31	J/molxK	725.45	Joback Method
cpg	456.89	J/molxK	766.03	Joback Method

cpg	471.80	J/mol×K	806.61	Joback Method
cpg	486.13	J/mol×K	847.19	Joback Method
hfust	24.00	kJ/mol	375.00	NIST Webbook
hfust	8.70	kJ/mol	495.00	NIST Webbook
sfust	64.00	J/mol×K	375.00	NIST Webbook
sfust	17.58	J/mol×K	495.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C595302&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C595302&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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