

# Naphthalene, 1,2,3,4-tetrahydro-5,8-dimethyl-1-octyl-

**Other names:** 1,2,3,4-Tetrahydro-5,8-dimethyl-1-octylnaphthalene

**Inchi:** InChI=1S/C20H32/c1-4-5-6-7-8-9-11-18-12-10-13-19-16(2)14-15-17(3)20(18)19/h14-15,1

**InchiKey:** YGXXXFDZLBVJPP-UHFFFAOYSA-N

**Formula:** C20H32

**SMILES:** CCCCCCCC1CCCc2c(C)ccc(C)c21

**Mol. weight [g/mol]:** 272.47

**CAS:** 55255-58-8

## Physical Properties

Property code	Value	Unit	Source
gf	249.69	kJ/mol	Joback Method
hf	-187.37	kJ/mol	Joback Method
hfus	36.46	kJ/mol	Joback Method
hvap	64.46	kJ/mol	Joback Method
log10ws	-7.24		Crippen Method
logp	6.474		Crippen Method
mcvol	258.040	ml/mol	McGowan Method
pc	1374.80	kPa	Joback Method
tb	709.63	K	Joback Method
tc	909.89	K	Joback Method
tf	393.56	K	Joback Method
vc	0.997	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	866.81	J/molxK	909.89	Joback Method
cpg	851.09	J/molxK	876.52	Joback Method
cpg	834.43	J/molxK	843.14	Joback Method
cpg	816.79	J/molxK	809.76	Joback Method
cpg	798.11	J/molxK	776.38	Joback Method
cpg	778.33	J/molxK	743.01	Joback Method
cpg	757.39	J/molxK	709.63	Joback Method
dvisc	0.0014387	Paxs	393.56	Joback Method

dvisc	0.0001939	Paxs	709.63	Joback Method
dvisc	0.0002369	Paxs	656.95	Joback Method
dvisc	0.0002996	Paxs	604.27	Joback Method
dvisc	0.0003964	Paxs	551.60	Joback Method
dvisc	0.0005563	Paxs	498.92	Joback Method
dvisc	0.0008459	Paxs	446.24	Joback Method
hvapt	78.60	kJ/mol	450.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55255588&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55255588&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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