

# 17-Methyl-16,17-dihydro-15H-cyclopenta(a)phenanthrene

<b>Inchi:</b>	InChI=1S/C18H16/c1-12-6-8-16-14(12)10-11-17-15-5-3-2-4-13(15)7-9-18(16)17/h2-5,7,9
<b>InchiKey:</b>	GRGSXXHCCGGRX-UHFFFAOYSA-N
<b>Formula:</b>	C18H16
<b>SMILES:</b>	CC1CCc2c1ccc1c2ccc2ccccc21
<b>Mol. weight [g/mol]:</b>	232.32

## Physical Properties

Property code	Value	Unit	Source
gf	458.25	kJ/mol	Joback Method
hf	242.21	kJ/mol	Joback Method
hfus	27.42	kJ/mol	Joback Method
hvap	63.12	kJ/mol	Joback Method
log10ws	-6.54		Crippen Method
logp	5.043		Crippen Method
mvol	190.940	ml/mol	McGowan Method
pc	2402.92	kPa	Joback Method
rinpol	375.72		NIST Webbook
rinpol	374.63		NIST Webbook
tb	697.56	K	Joback Method
tc	948.16	K	Joback Method
tf	439.94	K	Joback Method
vc	0.737	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	516.99	J/molxK	697.56	Joback Method
cpg	590.94	J/molxK	906.40	Joback Method
cpg	577.92	J/molxK	864.63	Joback Method
cpg	564.20	J/molxK	822.86	Joback Method
cpg	549.59	J/molxK	781.09	Joback Method
cpg	533.91	J/molxK	739.33	Joback Method
cpg	603.44	J/molxK	948.16	Joback Method
dvisc	0.0010570	Paxs	697.56	Joback Method

dvisc	0.0011368	Paxs	654.62	Joback Method
dvisc	0.0012352	Paxs	611.69	Joback Method
dvisc	0.0013590	Paxs	568.75	Joback Method
dvisc	0.0015188	Paxs	525.81	Joback Method
dvisc	0.0017312	Paxs	482.88	Joback Method
dvisc	0.0020243	Paxs	439.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R214483&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R214483&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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