

# 1H-Indene, 2-butyl-5-hexyloctahydro-

<b>Other names:</b>	Bicyclo[4.3.0]nonane, 8-butyl-3-hexyl-2-n-Butyl-5-n-hexyl-(hexahydroindan) 8-butyl-3-hexylbicyclo[4.3.0]nonane
<b>Inchi:</b>	InChI=1S/C19H36/c1-3-5-7-8-10-16-11-12-18-14-17(9-6-4-2)15-19(18)13-16/h16-19H,3-
<b>InchiKey:</b>	FLHBJKDPWVFHDH-UHFFFAOYSA-N
<b>Formula:</b>	C19H36
<b>SMILES:</b>	CCCCCCC1CCC2CC(CCCC)CC2C1
<b>Mol. weight [g/mol]:</b>	264.49
<b>CAS:</b>	55044-33-2

## Physical Properties

Property code	Value	Unit	Source
gf	178.88	kJ/mol	Joback Method
hf	-349.05	kJ/mol	Joback Method
hfus	37.08	kJ/mol	Joback Method
hvap	57.61	kJ/mol	Joback Method
log10ws	-6.60		Crippen Method
logp	6.590		Crippen Method
mcvol	256.850	ml/mol	McGowan Method
pc	1300.47	kPa	Joback Method
tb	651.07	K	Joback Method
tc	840.02	K	Joback Method
tf	320.73	K	Joback Method
vc	0.988	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	767.61	J/molxK	651.07	Joback Method
cpg	878.76	J/molxK	808.53	Joback Method
cpg	858.92	J/molxK	777.03	Joback Method
cpg	837.93	J/molxK	745.54	Joback Method
cpg	815.75	J/molxK	714.05	Joback Method
cpg	792.33	J/molxK	682.56	Joback Method

cpg	897.51	J/mol×K	840.02	Joback Method
dvisc	0.0004189	Paxs	651.07	Joback Method
dvisc	0.0004976	Paxs	596.01	Joback Method
dvisc	0.0006120	Paxs	540.96	Joback Method
dvisc	0.0007890	Paxs	485.90	Joback Method
dvisc	0.0010854	Paxs	430.84	Joback Method
dvisc	0.0016392	Paxs	375.79	Joback Method
dvisc	0.0028523	Paxs	320.73	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55044332&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55044332&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/76-075-5/1H-Indene-2-butyl-5-hexyloctahydro.pdf>

Generated by Cheméo on 2024-04-19 15:46:15.519625722 +0000 UTC m=+15830824.440203038.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.