

BENZENEBUTANOL

Other names:	1-butanol, 4-phenyl- 4-Phenyl butanol-1 4-Phenyl-n-butanol 4-Phenylbutanol 4-phenyl-1-butanol 4-phenylbutan-1-ol Phenylbutyl alcohol
Inchi:	InChI=1S/C10H14O/c11-9-5-4-8-10-6-2-1-3-7-10/h1-3,6-7,11H,4-5,8-9H2
InchiKey:	LDZLXQFDGRCELX-UHFFFAOYSA-N
Formula:	C10H14O
SMILES:	OCCCCc1ccccc1
Mol. weight [g/mol]:	150.22

Physical Properties

Property code	Value	Unit	Source
af	0.6853		Cheméo Relay (1.0)
gf	8.91	kJ/mol	Joback Method
hf	-159.85	kJ/mol	Cheméo Relay (1.0)
hfus	19.79	kJ/mol	Joback Method
hvap	83.75	kJ/mol	Cheméo Relay (1.0)
ie	8.81	eV	Cheméo Relay (1.0)
log10ws	-1.70		Cheméo Relay (1.0)
logp	2.002		Crippen Method
mcvol	133.870	ml/mol	McGowan Method
pc	3070.00	kPa	Vapor-Liquid Critical Properties of Phenol and (C8 to C10) Phenylalkanols
tb	539.60	K	Cheméo Relay (1.0)
tc	756.86	K	Cheméo Relay (1.0)
tf	258.57	K	Cheméo Relay (1.0)
vc	0.483	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	311.72	J/mol×K	547.06	Joback Method
cpg	324.22	J/mol×K	579.13	Joback Method
cpg	336.04	J/mol×K	611.20	Joback Method
cpg	347.19	J/mol×K	643.27	Joback Method
cpg	357.73	J/mol×K	675.33	Joback Method
cpg	367.66	J/mol×K	707.40	Joback Method
cpg	377.01	J/mol×K	739.47	Joback Method
dvisc	0.0140109	Pa×s	289.70	Joback Method
dvisc	0.0036769	Pa×s	332.59	Joback Method
dvisc	0.0013099	Pa×s	375.49	Joback Method
dvisc	0.0005766	Pa×s	418.38	Joback Method
dvisc	0.0002957	Pa×s	461.27	Joback Method
dvisc	0.0001699	Pa×s	504.17	Joback Method
dvisc	0.0001065	Pa×s	547.06	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	413.20	K	1.90	NIST Webbook

Sources

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3360416&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Vapor-Liquid Critical Properties of Phenol and (C8 to C10) Phenylalkanols: Partitioning of Phenylalkanols between Micelles and Water Studied by Limiting Interdiffusion Coefficients in Water and Tetradecyltrimethylammonium Bromide Solutions:

<https://www.doi.org/10.1021/je0604380>

<https://www.doi.org/10.1021/je900967j>

Legend

af:	Acentric Factor
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/76-083-6/BENZENEBUTANOL.pdf>

Generated by Cheméo on 2026-07-28 03:35:42.675988314 +0000 UTC m=+75500.092702266.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.