

# Bicycloheptyl-3,3'-diene

<b>Inchi:</b>	InChI=1S/C14H22/c1-2-6-10-13(9-5-1)14-11-7-3-4-8-12-14/h1,3,5,7,13-14H,2,4,6,8-12H2
<b>InchiKey:</b>	KMYUSDGASPTLGY-UHFFFAOYSA-N
<b>Formula:</b>	C14H22
<b>SMILES:</b>	C1=CCC(C2CC=CCCC2)CCC1
<b>Mol. weight [g/mol]:</b>	190.32

## Physical Properties

Property code	Value	Unit	Source
gf	151.62	kJ/mol	Joback Method
hf	-120.41	kJ/mol	Joback Method
hfus	13.93	kJ/mol	Joback Method
hvap	48.54	kJ/mol	Joback Method
log10ws	-4.70		Crippen Method
logp	4.479		Crippen Method
mcvol	177.800	ml/mol	McGowan Method
pc	2426.65	kPa	Joback Method
rinsol	1556.00		NIST Webbook
tb	565.68	K	Joback Method
tc	810.92	K	Joback Method
tf	256.78	K	Joback Method
vc	0.641	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	454.90	J/molxK	565.68	Joback Method
cpg	481.78	J/molxK	606.55	Joback Method
cpg	506.80	J/molxK	647.43	Joback Method
cpg	530.01	J/molxK	688.30	Joback Method
cpg	551.45	J/molxK	729.18	Joback Method
cpg	571.19	J/molxK	770.05	Joback Method
cpg	589.26	J/molxK	810.92	Joback Method
dvisc	0.0120754	Paxs	256.78	Joback Method
dvisc	0.0030564	Paxs	308.26	Joback Method

dvisc	0.0011463	Paxs	359.75	Joback Method
dvisc	0.0005496	Paxs	411.23	Joback Method
dvisc	0.0003103	Paxs	462.71	Joback Method
dvisc	0.0001965	Paxs	514.20	Joback Method
dvisc	0.0001352	Paxs	565.68	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R136457&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R136457&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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