

# Benzene, 1-butyl-2,4-dimethyl

<b>Other names:</b>	1,3-Dimethyl-4-n-Butylbenzene
<b>Inchi:</b>	InChI=1S/C12H18/c1-4-5-6-12-8-7-10(2)9-11(12)3/h7-9H,4-6H2,1-3H3
<b>InchiKey:</b>	JEHCWPAYIBNVDI-UHFFFAOYSA-N
<b>Formula:</b>	C12H18
<b>SMILES:</b>	CCCCc1ccc(C)cc1C
<b>Mol. weight [g/mol]:</b>	162.27

## Physical Properties

Property code	Value	Unit	Source
gf	143.31	kJ/mol	Joback Method
hf	-77.42	kJ/mol	Joback Method
hfus	20.10	kJ/mol	Joback Method
hvap	45.91	kJ/mol	Joback Method
log10ws	-4.06		Crippen Method
logp	3.646		Crippen Method
mcvol	156.180	ml/mol	McGowan Method
pc	2336.03	kPa	Joback Method
rinpol	1249.00		NIST Webbook
rinpol	1249.00		NIST Webbook
rinpol	1250.00		NIST Webbook
rinpol	1249.00		NIST Webbook
ripol	1487.00		NIST Webbook
ripol	1487.00		NIST Webbook
ripol	1487.30		NIST Webbook
ripol	1487.00		NIST Webbook
tb	510.60	K	Joback Method
tc	712.18	K	Joback Method
tf	276.46	K	Joback Method
vc	0.600	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	346.80	J/mol×K	510.60	Joback Method

cpg	362.94	J/molxK	544.20	Joback Method
cpg	378.30	J/molxK	577.79	Joback Method
cpg	392.88	J/molxK	611.39	Joback Method
cpg	406.73	J/molxK	644.99	Joback Method
cpg	419.85	J/molxK	678.59	Joback Method
cpg	432.28	J/molxK	712.18	Joback Method
dvisc	0.0018797	Paxs	276.46	Joback Method
dvisc	0.0010177	Paxs	315.48	Joback Method
dvisc	0.0006307	Paxs	354.51	Joback Method
dvisc	0.0004298	Paxs	393.53	Joback Method
dvisc	0.0003139	Paxs	432.55	Joback Method
dvisc	0.0002414	Paxs	471.58	Joback Method
dvisc	0.0001933	Paxs	510.60	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R52939&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R52939&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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