

# Cyclohexane, (1,1-dimethylethyl)-

<b>Other names:</b>	(1,1-DIMETHYLETHYL)CYCLOHEXANE 1,1-dimethylethylcyclohexane 1-tert-Butylcyclohexane 2-cyclohexyl-2-methylpropane Cyclohexane, t-butyl- Cyclohexane, tert-butyl- t-Butylcyclohexane tert-Butylcyclohexane
<b>Inchi:</b>	InChI=1S/C10H20/c1-10(2,3)9-7-5-4-6-8-9/h9H,4-8H2,1-3H3
<b>InchiKey:</b>	XTVMZZBLCLWBPM-UHFFFAOYSA-N
<b>Formula:</b>	C10H20
<b>SMILES:</b>	CC(C)(C)C1CCCCC1
<b>Mol. weight [g/mol]:</b>	140.27
<b>CAS:</b>	3178-22-1

## Physical Properties

Property code	Value	Unit	Source
af	0.2520		KDB
dm	0.00	debye	KDB
gf	60.61	kJ/mol	Joback Method
hf	-204.16	kJ/mol	Joback Method
hfus	6.08	kJ/mol	Joback Method
hvap	46.98	kJ/mol	NIST Webbook
log10ws	-3.42		Crippen Method
logp	3.613		Crippen Method
mcvol	140.900	ml/mol	McGowan Method
pc	2660.00	kPa	KDB
rinpol	1025.00		NIST Webbook
rinpol	1012.00		NIST Webbook
rinpol	1021.00		NIST Webbook
rinpol	1003.00		NIST Webbook
rinpol	1005.00		NIST Webbook
rinpol	984.00		NIST Webbook
rinpol	1027.10		NIST Webbook
rinpol	978.10		NIST Webbook
rinpol	982.20		NIST Webbook
rinpol	985.20		NIST Webbook

rinpol	983.10		NIST Webbook
rinpol	1014.10		NIST Webbook
rinpol	1000.00		NIST Webbook
rinpol	1014.10		NIST Webbook
rinpol	1025.00		NIST Webbook
rinpol	1014.00		NIST Webbook
rinpol	988.00		NIST Webbook
ripol	1088.00		NIST Webbook
ripol	1062.00		NIST Webbook
ripol	1075.00		NIST Webbook
tb	444.70	K	NIST Webbook
tb	444.90 ± 0.50	K	NIST Webbook
tb	444.70	K	NIST Webbook
tb	444.74 ± 0.03	K	NIST Webbook
tb	440.20	K	NIST Webbook
tb	444.69 ± 0.20	K	NIST Webbook
tb	444.70	K	KDB
tc	659.00	K	NIST Webbook
tc	659.00	K	KDB
tc	652.00	K	Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons
tf	232.00	K	KDB
tf	231.95 ± 0.20	K	NIST Webbook
vc	0.517	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	396.64	J/molxK	619.69	Joback Method
cpg	301.81	J/molxK	444.52	Joback Method
cpg	323.14	J/molxK	479.55	Joback Method
cpg	343.23	J/molxK	514.59	Joback Method
cpg	362.15	J/molxK	549.62	Joback Method
cpg	379.94	J/molxK	584.66	Joback Method
cpg	412.31	J/molxK	654.72	Joback Method
cpl	264.80	J/molxK	313.00	NIST Webbook
dvisc	0.0002719	Paxs	444.52	Joback Method
dvisc	0.0145506	Paxs	212.26	Joback Method
dvisc	0.0044939	Paxs	250.97	Joback Method
dvisc	0.0019000	Paxs	289.68	Joback Method

dvisc	0.0009840	Paxs	328.39	Joback Method
dvisc	0.0005855	Paxs	367.10	Joback Method
dvisc	0.0003846	Paxs	405.81	Joback Method
hvapt	42.40 ± 0.10	kJ/mol	368.00	NIST Webbook
hvapt	37.01	kJ/mol	444.70	NIST Webbook
hvapt	42.90	kJ/mol	400.50	NIST Webbook
hvapt	45.00 ± 0.10	kJ/mol	328.00	NIST Webbook
hvapt	44.00 ± 0.10	kJ/mol	343.00	NIST Webbook
hvapt	43.00 ± 0.10	kJ/mol	358.00	NIST Webbook
rhol	813.00	kg/m3	293.00	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.32339e+01
Coeff. B	-3.15642e+03
Coeff. C	-7.76360e+01
Temperature range (K), min.	321.45
Temperature range (K), max.	476.05

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	8.63471e+01
Coeff. B	-8.58272e+03
Coeff. C	-1.03859e+01
Coeff. D	4.50963e-06
Temperature range (K), min.	273.15
Temperature range (K), max.	659.00

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

Gas-Liquid Critical Temperatures of Some Alkenes, Amines, and Cyclic Hydrocarbons, Fluorenone, thianthrene and xanthene in organic solvents: NIST Webbook:

<https://www.doi.org/10.1021/je0341357>

KDB Vapor Pressure Data:

<https://www.doi.org/10.1016/j.fluid.2005.02.016>

McGowan Method:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3178221&Units=SI>

KDB:

<https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=587>

The Yaws Handbook of Vapor Pressure:

<http://link.springer.com/article/10.1007/BF02311772>

<https://www.thermo.com/files/research/kdb/mol/mol587.mol>

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

## Legend

<b>af:</b>	Acentric Factor
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dm:</b>	Dipole Moment
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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