

# Benzene, (1-ethyl-2-propenyl)-

<b>Other names:</b>	1-Pentene, 3-phenyl- Benzene, (1-ethylallyl)- 3-Phenyl-1-pentene
<b>Inchi:</b>	InChI=1S/C11H14/c1-3-10(4-2)11-8-6-5-7-9-11/h3,5-10H,1,4H2,2H3
<b>InchiKey:</b>	HAGOWDKLLDRZAS-UHFFFAOYSA-N
<b>Formula:</b>	C11H14
<b>SMILES:</b>	C=CC(CC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	146.23
<b>CAS:</b>	19947-22-9

## Physical Properties

Property code	Value	Unit	Source
gf	239.55	kJ/mol	Joback Method
hf	86.31	kJ/mol	Joback Method
hfus	13.48	kJ/mol	Joback Method
hvap	41.30	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	3.366		Crippen Method
mcvol	137.790	ml/mol	McGowan Method
pc	2790.61	kPa	Joback Method
tb	474.00	K	Joback Method
tc	685.20	K	Joback Method
tf	223.39	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.66	J/mol×K	474.00	Joback Method
cpg	298.57	J/mol×K	509.20	Joback Method
cpg	313.54	J/mol×K	544.40	Joback Method
cpg	327.59	J/mol×K	579.60	Joback Method
cpg	340.77	J/mol×K	614.80	Joback Method
cpg	353.13	J/mol×K	650.00	Joback Method

cpg	364.71	J/mol×K	685.20	Joback Method
dvisc	0.0051264	Paxs	223.39	Joback Method
dvisc	0.0019756	Paxs	265.16	Joback Method
dvisc	0.0009869	Paxs	306.93	Joback Method
dvisc	0.0005822	Paxs	348.69	Joback Method
dvisc	0.0003845	Paxs	390.46	Joback Method
dvisc	0.0002751	Paxs	432.23	Joback Method
dvisc	0.0002089	Paxs	474.00	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19947229&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19947229&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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