

# Decane, 5-ethyl-5-methyl-

<b>Other names:</b>	5-ethyl-5-methyldecane
<b>Inchi:</b>	InChI=1S/C13H28/c1-5-8-10-12-13(4,7-3)11-9-6-2/h5-12H2,1-4H3
<b>InchiKey:</b>	METKCQZZZVWVCD-UHFFFAOYSA-N
<b>Formula:</b>	C13H28
<b>SMILES:</b>	CCCCC(C)(CC)CCCC
<b>Mol. weight [g/mol]:</b>	184.36
<b>CAS:</b>	17312-74-2

## Physical Properties

Property code	Value	Unit	Source
gf	61.42	kJ/mol	Joback Method
hf	-320.40	kJ/mol	Joback Method
hfus	22.01	kJ/mol	Joback Method
hvap	43.24	kJ/mol	Joback Method
log10ws	-5.02		Crippen Method
logp	5.173		Crippen Method
mcvol	194.030	ml/mol	McGowan Method
pc	1670.06	kPa	Joback Method
tb	493.61	K	Joback Method
tc	662.67	K	Joback Method
tf	238.69	K	Joback Method
vc	0.752	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	457.75	J/molxK	493.61	Joback Method
cpg	476.43	J/molxK	521.79	Joback Method
cpg	494.29	J/molxK	549.96	Joback Method
cpg	511.35	J/molxK	578.14	Joback Method
cpg	527.65	J/molxK	606.32	Joback Method
cpg	543.21	J/molxK	634.50	Joback Method
cpg	558.07	J/molxK	662.67	Joback Method
dvisc	0.0090102	Paxs	238.69	Joback Method

dvisc	0.0029290	Paxs	281.18	Joback Method
dvisc	0.0012789	Paxs	323.66	Joback Method
dvisc	0.0006768	Paxs	366.15	Joback Method
dvisc	0.0004088	Paxs	408.64	Joback Method
dvisc	0.0002716	Paxs	451.12	Joback Method
dvisc	0.0001935	Paxs	493.61	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17312742&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17312742&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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