Valeric anhydride

Other names:	Pentanoic acid, anhydride
	n-Valeric anhydride
	pentanoic anhydride
Inchi:	InChI=1S/C10H18O3/c1-3-5-7-9(11)13-10(12)8-6-4-2/h3-8H2,1-2H3
InchiKey:	DUCKXCGALKOSJF-UHFFFAOYSA-N
Formula:	C10H18O3
SMILES:	CCCCC(=O)OC(=O)CCCC
Mol. weight [g/mol]:	186.25
CAS:	2082-59-9

Physical Properties

Property code	Value	Unit	Source
gf	-329.52	kJ/mol	Joback Method
hf	-607.11	kJ/mol	Joback Method
hfus	26.04	kJ/mol	Joback Method
hvap	53.76	kJ/mol	Joback Method
log10ws	-2.65		Crippen Method
logp	2.437		Crippen Method
mcvol	160.770	ml/mol	McGowan Method
pc	2336.03	kPa	Joback Method
rinpol	1283.10		NIST Webbook
tb	558.36	K	Joback Method
tc	739.30	K	Joback Method
tf	324.55	K	Joback Method
vc	0.625	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	390.21	J/mol×K	558.36	Joback Method
cpg	403.56	J/mol×K	588.52	Joback Method
cpg	416.35	J/mol×K	618.67	Joback Method
cpg	428.57	J/mol×K	648.83	Joback Method
cpg	440.25	J/mol×K	678.99	Joback Method

cpg	451.38	J/mol×K	709.15	Joback Method
cpg	461.96	J/mol×K	739.30	Joback Method
dvisc	0.0026272	Pa×s	324.55	Joback Method
dvisc	0.0014182	Pa×s	363.52	Joback Method
dvisc	0.0008627	Pa×s	402.49	Joback Method
dvisc	0.0005729	Pa×s	441.46	Joback Method
dvisc	0.0004066	Pa×s	480.42	Joback Method
dvisc	0.0003038	Pa×s	519.39	Joback Method
dvisc	0.0002364	Pa×s	558.36	Joback Method
rhol	939.15	kg/m3	298.15	Excess Molar Enthalpies of Binary Systems of n-Valeric Anhydride or n-Hexanoic Anhydride with n-Dodecane, n-Tetradecane, or n-Hexadecane at 298.15 K

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2082599&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Excess Molar Enthalpies of Binary Systems of n-Valeric Anhydride or n-Hexanoic Anhydride with n-Dodecane, n-Tetradecane, or n-Hexadecane at 298.15 K:	https://www.doi.org/10.1021/je900598y
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure

rhol:	Liquid Density
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume