

8-hydroxy-p-menthan-3-one

Inchi:	InChI=1S/C10H18O2/c1-7-4-5-8(9(11)6-7)10(2,3)12/h7-8,12H,4-6H2,1-3H3
InchiKey:	QNZZGEDTDYWNFN-UHFFFAOYSA-N
Formula:	C10H18O2
SMILES:	CC1CCC(C(C)(C)O)C(=O)C1
Mol. weight [g/mol]:	170.25

Physical Properties

Property code	Value	Unit	Source
gf	-206.51	kJ/mol	Joback Method
hf	-514.43	kJ/mol	Joback Method
hfus	10.75	kJ/mol	Joback Method
hvap	57.60	kJ/mol	Joback Method
log10ws	-2.08		Crippen Method
logp	1.763		Crippen Method
mcvol	148.340	ml/mol	McGowan Method
pc	2890.51	kPa	Joback Method
rinpol	1231.00		NIST Webbook
rinpol	1231.00		NIST Webbook
tb	599.85	K	Joback Method
tc	810.28	K	Joback Method
tf	337.06	K	Joback Method
vc	0.542	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.95	J/mol×K	599.85	Joback Method
cpg	423.95	J/mol×K	634.92	Joback Method
cpg	439.99	J/mol×K	669.99	Joback Method
cpg	455.07	J/mol×K	705.07	Joback Method
cpg	469.22	J/mol×K	740.14	Joback Method
cpg	482.44	J/mol×K	775.21	Joback Method
cpg	494.75	J/mol×K	810.28	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R224316&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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