

# 1-Phenanthrenecarboxaldehyde, 1,2,3,4,4a,9,10,10a-octahydro-1,4a-dimethyl-7-(1-methyl-2-propenyl)-

Other names: Podocarpa-8,11,13-trien-16-al, 13-isopropyl-4-Epiabietal, dehydro-

(1S,4aS,10aR)-7-Isopropyl-1,4a-dimethyl-1,2,3,4,4a,9,10,10a-octahydrophenanthrene-1-

<b>Inchi:</b>	InChI=1S/C20H28O/c1-14(2)15-6-8-17-16(12-15)7-9-18-19(3,13-21)10-5-11-20(17,18)4/
<b>InchiKey:</b>	YCLCHPWGSDZKL-UHFFFAOYSA-N
<b>Formula:</b>	C20H28O
<b>SMILES:</b>	CC(C)c1ccc2c(c1)CCC1C(C)(C=O)CCCC21C
<b>Mol. weight [g/mol]:</b>	284.44
<b>CAS:</b>	24035-50-5

## Physical Properties

Property code	Value	Unit	Source
gf	187.32	kJ/mol	Joback Method
hf	-189.98	kJ/mol	Joback Method
hfus	20.13	kJ/mol	Joback Method
hvap	67.61	kJ/mol	Joback Method
log10ws	-5.45		Crippen Method
logp	5.019		Crippen Method
mcvol	248.750	ml/mol	McGowan Method
pc	1753.60	kPa	Joback Method
rinpol	2289.20		NIST Webbook
rinpol	2289.20		NIST Webbook
tb	759.69	K	Joback Method
tc	996.72	K	Joback Method
tf	466.02	K	Joback Method
vc	0.952	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	764.98	J/mol×K	759.69	Joback Method
cpg	787.50	J/mol×K	799.19	Joback Method
cpg	809.58	J/mol×K	838.70	Joback Method
cpg	831.56	J/mol×K	878.20	Joback Method

cpg	853.76	J/mol×K	917.71	Joback Method
cpg	876.53	J/mol×K	957.21	Joback Method
cpg	900.19	J/mol×K	996.72	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24035505&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24035505&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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