

# Cyclopentene,3-(3-methylbutyl)-

**Inchi:** InChI=1S/C10H18/c1-9(2)7-8-10-5-3-4-6-10/h3,5,9-10H,4,6-8H2,1-2H3  
**InchiKey:** BUCQADXVDGFUTQ-UHFFFAOYSA-N  
**Formula:** C10H18  
**SMILES:** CC(C)CCC1C=CCC1  
**Mol. weight [g/mol]:** 138.25  
**CAS:** 37689-16-0

## Physical Properties

Property code	Value	Unit	Source
gf	97.39	kJ/mol	Joback Method
hf	-136.75	kJ/mol	Joback Method
hfus	13.29	kJ/mol	Joback Method
hvap	38.02	kJ/mol	Joback Method
ie	8.83 ± 0.02	eV	NIST Webbook
log10ws	-3.27		Crippen Method
logp	3.389		Crippen Method
mcvol	136.600	ml/mol	McGowan Method
pc	2619.09	kPa	Joback Method
rinpol	975.70		NIST Webbook
rinpol	976.00		NIST Webbook
rinpol	972.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	978.00		NIST Webbook
rinpol	972.00		NIST Webbook
rinpol	978.00		NIST Webbook
ripol	1094.00		NIST Webbook
ripol	1094.00		NIST Webbook
ripol	1088.00		NIST Webbook
ripol	1094.00		NIST Webbook
ripol	1095.20		NIST Webbook
ripol	1081.70		NIST Webbook
ripol	1088.90		NIST Webbook
ripol	1095.20		NIST Webbook
ripol	1081.70		NIST Webbook
ripol	1088.90		NIST Webbook
ripol	1088.90		NIST Webbook
ripol	1095.20		NIST Webbook

ripol	1097.00		NIST Webbook
ripol	1101.00		NIST Webbook
tb	442.20	K	Joback Method
tc	637.86	K	Joback Method
tf	199.12	K	Joback Method
vc	0.516	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	285.57	J/mol×K	442.20	Joback Method
cpg	303.36	J/mol×K	474.81	Joback Method
cpg	320.25	J/mol×K	507.42	Joback Method
cpg	336.28	J/mol×K	540.03	Joback Method
cpg	351.46	J/mol×K	572.64	Joback Method
cpg	365.85	J/mol×K	605.25	Joback Method
cpg	379.45	J/mol×K	637.86	Joback Method
dvisc	0.0059671	Paxs	199.12	Joback Method
dvisc	0.0023351	Paxs	239.63	Joback Method
dvisc	0.0011987	Paxs	280.15	Joback Method
dvisc	0.0007282	Paxs	320.66	Joback Method
dvisc	0.0004948	Paxs	361.17	Joback Method
dvisc	0.0003634	Paxs	401.69	Joback Method
dvisc	0.0002824	Paxs	442.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37689160&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37689160&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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