

# m-Menth-6-ene, (R)-(+)-

<b>Other names:</b>	m-Menth-6-ene
<b>Inchi:</b>	InChI=1S/C10H18/c1-8(2)10-6-4-5-9(3)7-10/h5,8,10H,4,6-7H2,1-3H3/t10-/m1/s1
<b>InchiKey:</b>	XLXFKTBODPBMN-SNVBAGLBSA-N
<b>Formula:</b>	C10H18
<b>SMILES:</b>	CC1=CCCC(C(C)C)C1
<b>Mol. weight [g/mol]:</b>	138.25
<b>CAS:</b>	13837-70-2

## Physical Properties

Property code	Value	Unit	Source
gf	75.66	kJ/mol	Joback Method
hf	-154.38	kJ/mol	Joback Method
hfus	10.80	kJ/mol	Joback Method
hvap	38.85	kJ/mol	Joback Method
log10ws	-3.27		Crippen Method
logp	3.389		Crippen Method
mcvol	136.600	ml/mol	McGowan Method
pc	2646.11	kPa	Joback Method
rinpol	947.00		NIST Webbook
rinpol	947.00		NIST Webbook
tb	451.45	K	Joback Method
tc	656.32	K	Joback Method
tf	208.12	K	Joback Method
vc	0.508	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	286.78	J/mol×K	451.45	Joback Method
cpg	305.15	J/mol×K	485.59	Joback Method
cpg	322.63	J/mol×K	519.74	Joback Method
cpg	339.23	J/mol×K	553.88	Joback Method
cpg	354.99	J/mol×K	588.03	Joback Method
cpg	369.91	J/mol×K	622.17	Joback Method

cpg	384.04	J/mol×K	656.32	Joback Method
dvisc	0.0061732	Paxs	208.12	Joback Method
dvisc	0.0022871	Paxs	248.67	Joback Method
dvisc	0.0011194	Paxs	289.23	Joback Method
dvisc	0.0006532	Paxs	329.78	Joback Method
dvisc	0.0004288	Paxs	370.34	Joback Method
dvisc	0.0003059	Paxs	410.89	Joback Method
dvisc	0.0002319	Paxs	451.45	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13837702&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13837702&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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