

# Pentane, 2,2'-oxybis-

<b>Other names:</b>	bis(1-methylbutyl) ether
<b>Inchi:</b>	InChI=1S/C10H22O/c1-5-7-9(3)11-10(4)8-6-2/h9-10H,5-8H2,1-4H3
<b>InchiKey:</b>	UVEFRWVGQRNNDB-UHFFFAOYSA-N
<b>Formula:</b>	C10H22O
<b>SMILES:</b>	CCCC(C)OC(C)CCC
<b>Mol. weight [g/mol]:</b>	158.28
<b>CAS:</b>	56762-00-6

## Physical Properties

Property code	Value	Unit	Source
gf	-76.56	kJ/mol	Joback Method
hf	-392.51	kJ/mol	Joback Method
hfus	15.80	kJ/mol	Joback Method
hvap	39.49	kJ/mol	Joback Method
log10ws	-3.32		Crippen Method
logp	3.380		Crippen Method
mcvol	157.630	ml/mol	McGowan Method
pc	2106.13	kPa	Joback Method
tb	449.74	K	Joback Method
tc	619.25	K	Joback Method
tf	194.69	K	Joback Method
vc	0.602	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.48	J/molxK	449.74	Joback Method
cpg	414.87	J/molxK	591.00	Joback Method
cpg	401.45	J/molxK	562.75	Joback Method
cpg	387.51	J/molxK	534.50	Joback Method
cpg	373.04	J/molxK	506.24	Joback Method
cpg	358.04	J/molxK	477.99	Joback Method
cpg	427.77	J/molxK	619.25	Joback Method
dvisc	0.0001845	Paxs	449.74	Joback Method

dvisc	0.0002583	Paxs	407.23	Joback Method
dvisc	0.0003911	Paxs	364.72	Joback Method
dvisc	0.0006607	Paxs	322.22	Joback Method
dvisc	0.0013091	Paxs	279.71	Joback Method
dvisc	0.0033142	Paxs	237.20	Joback Method
dvisc	0.0125876	Paxs	194.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56762006&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56762006&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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