

# n-Propyl heptyl ether

<b>Other names:</b>	Heptyl propyl ether
<b>Inchi:</b>	InChI=1S/C10H22O/c1-3-5-6-7-8-10-11-9-4-2/h3-10H2,1-2H3
<b>InchiKey:</b>	DOYXDCAHTUMXDF-UHFFFAOYSA-N
<b>Formula:</b>	C10H22O
<b>SMILES:</b>	CCCCCCCCOCCC
<b>Mol. weight [g/mol]:</b>	158.28
<b>CAS:</b>	71112-89-5

## Physical Properties

Property code	Value	Unit	Source
gf	-71.68	kJ/mol	Joback Method
hf	-381.95	kJ/mol	Joback Method
hfus	22.84	kJ/mol	Joback Method
hvap	40.26	kJ/mol	Joback Method
log10ws	-3.10		Crippen Method
logp	3.383		Crippen Method
mcvol	157.630	ml/mol	McGowan Method
pc	2075.54	kPa	Joback Method
rinpol	1075.00		NIST Webbook
rinpol	1075.00		NIST Webbook
tb	450.62	K	Joback Method
tc	613.22	K	Joback Method
tf	224.69	K	Joback Method
vc	0.614	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.40	J/molxK	450.62	Joback Method
cpg	411.42	J/molxK	586.12	Joback Method
cpg	398.57	J/molxK	559.02	Joback Method
cpg	385.25	J/molxK	531.92	Joback Method
cpg	371.45	J/molxK	504.82	Joback Method
cpg	357.17	J/molxK	477.72	Joback Method

cpg	423.80	J/mol×K	613.22	Joback Method
dvisc	0.0002026	Paxs	450.62	Joback Method
dvisc	0.0002676	Paxs	412.97	Joback Method
dvisc	0.0003736	Paxs	375.31	Joback Method
dvisc	0.0005620	Paxs	337.65	Joback Method
dvisc	0.0009367	Paxs	300.00	Joback Method
dvisc	0.0018076	Paxs	262.35	Joback Method
dvisc	0.0043485	Paxs	224.69	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C71112895&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C71112895&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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