

# Octane, 3,6-dimethyl-

Other names:	3,6-Dimethyloctane
Inchi:	InChI=1S/C10H22/c1-5-9(3)7-8-10(4)6-2/h9-10H,5-8H2,1-4H3
InchiKey:	JEEQUUSFXRPRK-UHFFFAOYSA-N
Formula:	C10H22
SMILES:	CCC(C)CCC(C)CC
Mol. weight [g/mol]:	142.28
CAS:	15869-94-0

## Physical Properties

Property code	Value	Unit	Source
gf	28.44	kJ/mol	Joback Method
hf	-260.29	kJ/mol	Joback Method
hfus	14.61	kJ/mol	Joback Method
hvap	47.30	kJ/mol	NIST Webbook
log10ws	-3.52		Crippen Method
logp	3.859		Crippen Method
mcvol	151.760	ml/mol	McGowan Method
pc	2139.38	kPa	Joback Method
rinpol	929.00		NIST Webbook
rinpol	940.00		NIST Webbook
rinpol	940.00		NIST Webbook
rinpol	940.60		NIST Webbook
rinpol	940.60		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	927.80		NIST Webbook
rinpol	927.80		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	938.00		NIST Webbook
rinpol	948.30		NIST Webbook
rinpol	942.00		NIST Webbook
rinpol	944.00		NIST Webbook
rinpol	937.00		NIST Webbook
rinpol	944.00		NIST Webbook
rinpol	929.00		NIST Webbook
rinpol	942.00		NIST Webbook
rinpol	944.00		NIST Webbook

rinpol	948.30		NIST Webbook
tb	427.32	K	Joback Method
tc	596.92	K	Joback Method
tf	172.46	K	Joback Method
vc	0.584	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	315.26	J/molxK	427.32	Joback Method
cpg	331.18	J/molxK	455.59	Joback Method
cpg	346.50	J/molxK	483.85	Joback Method
cpg	361.21	J/molxK	512.12	Joback Method
cpg	375.35	J/molxK	540.39	Joback Method
cpg	388.93	J/molxK	568.65	Joback Method
cpg	401.95	J/molxK	596.92	Joback Method
dvisc	0.0214146	Paxs	172.46	Joback Method
dvisc	0.0046771	Paxs	214.94	Joback Method
dvisc	0.0016877	Paxs	257.41	Joback Method
dvisc	0.0008129	Paxs	299.89	Joback Method
dvisc	0.0004693	Paxs	342.37	Joback Method
dvisc	0.0003059	Paxs	384.84	Joback Method
dvisc	0.0002171	Paxs	427.32	Joback Method
rhol	746.60	kg/m3	283.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhol	738.90	kg/m3	293.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels

rhoI	731.10	kg/m3	303.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	723.30	kg/m3	313.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	715.50	kg/m3	323.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	707.60	kg/m3	333.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	699.60	kg/m3	343.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	691.50	kg/m3	353.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	683.30	kg/m3	363.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels

rhoI	674.90	kg/m3	373.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	746.60	kg/m3	283.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	738.90	kg/m3	293.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	731.20	kg/m3	303.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	723.50	kg/m3	313.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels
rhoI	715.70	kg/m3	323.15	Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels

Correlations

Information

Value

Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.41976e+01
Coeff. B	-3.53609e+03
Coeff. C	-6.48190e+01
Temperature range (K), min.	319.03
Temperature range (K), max.	462.75

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.06372e+02
Coeff. B	-9.38170e+03
Coeff. C	-1.34381e+01
Coeff. D	7.80491e-06
Temperature range (K), min.	319.15
Temperature range (K), max.	608.30

## Sources

The Yaws Handbook of Vapor Pressure:  
KDB Vapor Pressure Data:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

NIST Webbook:

<https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=113>

KDB:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C15869940&Units=SI>

Density, Viscosity, Speed of Sound, and Bulk Modulus of Methyl Alkanes, Dimethyl Alkanes, and Hydrotreated Renewable Fuels:  
McGowan Method:

<https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=113>

<https://www.doi.org/10.1021/je400274f>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307i>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/76-808-1/Octane-3-6-dimethyl.pdf>

Generated by Cheméo on 2025-12-22 06:00:04.189525959 +0000 UTC m=+6131401.719566623.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.