

5-Nonanone, 2-methyl-

Other names:	2-Methyl-5-nonanone
Inchi:	InChI=1S/C10H20O/c1-4-5-6-10(11)8-7-9(2)3/h9H,4-8H2,1-3H3
InchiKey:	YITHWSJQOVKDGI-UHFFFAOYSA-N
Formula:	C10H20O
SMILES:	CCCCC(=O)CCC(C)C
Mol. weight [g/mol]:	156.27
CAS:	22287-02-1

Physical Properties

Property code	Value	Unit	Source
gf	-98.04	kJ/mol	Joback Method
hf	-367.59	kJ/mol	Joback Method
hfus	19.73	kJ/mol	Joback Method
hvap	44.21	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	3.182		Crippen Method
mcvol	153.330	ml/mol	McGowan Method
pc	2252.53	kPa	Joback Method
tb	476.70	K	NIST Webbook
tb	477.00 ± 4.00	K	NIST Webbook
tc	658.01	K	Joback Method
tf	237.39	K	Joback Method
vc	0.596	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	342.14	J/mol×K	481.63	Joback Method
cpg	356.94	J/mol×K	511.03	Joback Method
cpg	371.13	J/mol×K	540.42	Joback Method
cpg	384.72	J/mol×K	569.82	Joback Method
cpg	397.72	J/mol×K	599.22	Joback Method
cpg	410.16	J/mol×K	628.61	Joback Method
cpg	422.05	J/mol×K	658.01	Joback Method

dvisc	0.0067728	Paxs	237.39	Joback Method
dvisc	0.0026268	Paxs	278.10	Joback Method
dvisc	0.0012976	Paxs	318.80	Joback Method
dvisc	0.0007520	Paxs	359.51	Joback Method
dvisc	0.0004869	Paxs	400.22	Joback Method
dvisc	0.0003417	Paxs	440.92	Joback Method
dvisc	0.0002545	Paxs	481.63	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.50947e+01
Coeff. B	-4.22242e+03
Coeff. C	-7.36590e+01
Temperature range (K), min.	358.82
Temperature range (K), max.	505.26

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C22287021&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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