

# 7-Octen-4-ol, 2,6-dimethyl-

<b>Other names:</b>	2,6-dimethyloct-7-en-4-ol
<b>Inchi:</b>	InChI=1S/C10H20O/c1-5-9(4)7-10(11)6-8(2)3/h5,8-11H,1,6-7H2,2-4H3
<b>InchiKey:</b>	GXRZVRGDJQMNEL-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O
<b>SMILES:</b>	C=CC(C)CC(O)CC(C)C
<b>Mol. weight [g/mol]:</b>	156.27
<b>CAS:</b>	94201-76-0

## Physical Properties

Property code	Value	Unit	Source
gf	-22.98	kJ/mol	Joback Method
hf	-292.37	kJ/mol	Joback Method
hfus	13.89	kJ/mol	Joback Method
hvap	52.70	kJ/mol	Joback Method
log10ws	-2.75		Crippen Method
logp	2.606		Crippen Method
mcvol	153.330	ml/mol	McGowan Method
pc	2450.74	kPa	Joback Method
rinpol	1082.00		NIST Webbook
tb	515.74	K	Joback Method
tc	685.98	K	Joback Method
tf	216.52	K	Joback Method
vc	0.578	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	362.56	J/molxK	515.74	Joback Method
cpg	376.18	J/molxK	544.11	Joback Method
cpg	389.21	J/molxK	572.49	Joback Method
cpg	401.68	J/molxK	600.86	Joback Method
cpg	413.59	J/molxK	629.23	Joback Method
cpg	424.97	J/molxK	657.60	Joback Method
cpg	435.83	J/molxK	685.98	Joback Method

dvisc	0.3552828	Paxs	216.52	Joback Method
dvisc	0.0260973	Paxs	266.39	Joback Method
dvisc	0.0043676	Paxs	316.26	Joback Method
dvisc	0.0011896	Paxs	366.13	Joback Method
dvisc	0.0004425	Paxs	416.00	Joback Method
dvisc	0.0002035	Paxs	465.87	Joback Method
dvisc	0.0001087	Paxs	515.74	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C94201760&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C94201760&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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