

# 9-Decen-1-ol

<b>Other names:</b>	1-Decen-10-ol 9-Decenol Dec-9-en-1-ol Decylenic alcohol «omega»-Decen-1-ol «omega»-Decenol Â«omegaÂ»-Decen-1-ol Â«omegaÂ»-Decenol
<b>Inchi:</b>	InChI=1S/C10H20O/c1-2-3-4-5-6-7-8-9-10-11/h2,11H,1,3-10H2
<b>InchiKey:</b>	QGFSQVPRCWJZQK-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O
<b>SMILES:</b>	C=CCCCCCCCCO
<b>Mol. weight [g/mol]:</b>	156.27
<b>CAS:</b>	13019-22-2

## Physical Properties

Property code	Value	Unit	Source
gf	-15.66	kJ/mol	Joback Method
hf	-276.53	kJ/mol	Joback Method
hfus	24.46	kJ/mol	Joback Method
hvap	53.86	kJ/mol	Joback Method
log10ws	-3.13		Crippen Method
logp	2.895		Crippen Method
mcvol	153.330	ml/mol	McGowan Method
pc	2393.53	kPa	Joback Method
rinpol	1251.00		NIST Webbook
rinpol	1262.00		NIST Webbook
rinpol	1267.00		NIST Webbook
rinpol	1243.90		NIST Webbook
rinpol	1243.00		NIST Webbook
rinpol	1256.00		NIST Webbook
rinpol	1230.00		NIST Webbook
rinpol	1243.00		NIST Webbook
ripol	1828.00		NIST Webbook
ripol	1775.00		NIST Webbook
ripol	1813.00		NIST Webbook
tb	509.20	K	NIST Webbook

tc	678.10	K	Joback Method
tf	261.52	K	Joback Method
vc	0.596	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	361.84	J/mol×K	517.06	Joback Method
cpg	374.55	J/mol×K	543.90	Joback Method
cpg	386.74	J/mol×K	570.74	Joback Method
cpg	398.45	J/mol×K	597.58	Joback Method
cpg	409.67	J/mol×K	624.42	Joback Method
cpg	420.43	J/mol×K	651.26	Joback Method
cpg	430.75	J/mol×K	678.10	Joback Method
dvisc	0.0312210	Paxs	261.52	Joback Method
dvisc	0.0065839	Paxs	304.11	Joback Method
dvisc	0.0020352	Paxs	346.70	Joback Method
dvisc	0.0008134	Paxs	389.29	Joback Method
dvisc	0.0003895	Paxs	431.88	Joback Method
dvisc	0.0002129	Paxs	474.47	Joback Method
dvisc	0.0001285	Paxs	517.06	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.58090e+01
Coeff. B	-4.77600e+03
Coeff. C	-8.24160e+01
Temperature range (K), min.	390.12
Temperature range (K), max.	537.38

## Sources

**The Yaws Handbook of Vapor**

**Pressure:**

**Crippen Method:**

**Crippen Method:**

**Joback Method:**

**McGowan Method:**

**NIST Webbook:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C13019222&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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