

# Cyclopentane, 1,2,3,4,5-pentamethoxy-, stereoisomer

<b>Other names:</b>	1,2,3,4,5-Pentamethoxycyclopentane
<b>Inchi:</b>	InChI=1S/C10H20O5/c1-11-6-7(12-2)9(14-4)10(15-5)8(6)13-3/h6-10H,1-5H3
<b>InchiKey:</b>	FNIWXGXPELOQSR-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O5
<b>SMILES:</b>	COC1C(OC)C(OC)C(OC)C1OC
<b>Mol. weight [g/mol]:</b>	220.26
<b>CAS:</b>	29887-59-0

## Physical Properties

Property code	Value	Unit	Source
gf	-485.97	kJ/mol	Joback Method
hf	-931.71	kJ/mol	Joback Method
hfus	25.82	kJ/mol	Joback Method
hvap	48.92	kJ/mol	Joback Method
log10ws	0.10		Crippen Method
logp	0.075		Crippen Method
mcvol	170.250	ml/mol	McGowan Method
pc	2038.23	kPa	Joback Method
tb	536.90	K	Joback Method
tc	720.17	K	Joback Method
tf	307.55	K	Joback Method
vc	0.623	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	440.98	J/mol×K	536.90	Joback Method
cpg	525.01	J/mol×K	689.63	Joback Method
cpg	509.40	J/mol×K	659.08	Joback Method
cpg	493.16	J/mol×K	628.54	Joback Method
cpg	476.31	J/mol×K	597.99	Joback Method
cpg	458.90	J/mol×K	567.45	Joback Method
cpg	539.92	J/mol×K	720.17	Joback Method
dvisc	0.0001806	Paxs	536.90	Joback Method

dvisc	0.0002019	Paxs	498.67	Joback Method
dvisc	0.0002300	Paxs	460.45	Joback Method
dvisc	0.0002683	Paxs	422.23	Joback Method
dvisc	0.0003226	Paxs	384.00	Joback Method
dvisc	0.0004041	Paxs	345.77	Joback Method
dvisc	0.0005354	Paxs	307.55	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C29887590&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C29887590&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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