

# 1,4-Naphthalenedione, 2,3-dihydroxy-

<b>Other names:</b>	1,4-Naphthoquinone, 2,3-dihydroxy- Isonaphthazarin Isonaphthazarine 2,3-Dihydroxy-1,4-naphthoquinone 2,3-dihydroxy-1,4-naphthochinon
<b>Inchi:</b>	InChI=1S/C10H6O4/c11-7-5-3-1-2-4-6(5)8(12)10(14)9(7)13/h1-4,13-14H
<b>InchiKey:</b>	BVQUETZBXIMAFZ-UHFFFAOYSA-N
<b>Formula:</b>	C10H6O4
<b>SMILES:</b>	O=C1C(O)=C(O)C(=O)c2ccccc21
<b>Mol. weight [g/mol]:</b>	190.15
<b>CAS:</b>	605-37-8

## Physical Properties

Property code	Value	Unit	Source
gf	-315.66	kJ/mol	Joback Method
hf	-482.71	kJ/mol	Joback Method
hfus	17.91	kJ/mol	Joback Method
hvap	84.65	kJ/mol	Joback Method
log10ws	-2.18		Crippen Method
logp	1.393		Crippen Method
mcvol	127.720	ml/mol	McGowan Method
pc	5001.51	kPa	Joback Method
rinpol	1241.00		NIST Webbook
tb	804.66	K	Joback Method
tc	1029.44	K	Joback Method
tf	543.94	K	Joback Method
vc	0.475	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	352.58	J/molxK	804.66	Joback Method
cpg	360.68	J/molxK	842.12	Joback Method
cpg	368.06	J/molxK	879.59	Joback Method

cpg	374.71	J/mol×K	917.05	Joback Method
cpg	380.61	J/mol×K	954.51	Joback Method
cpg	385.75	J/mol×K	991.97	Joback Method
cpg	390.11	J/mol×K	1029.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C605378&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C605378&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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