

# 2,3-Dibromonaphthalene

<b>Other names:</b>	Naphthalene, 2,3-dibromo-
<b>Inchi:</b>	InChI=1S/C10H6Br2/c11-9-5-7-3-1-2-4-8(7)6-10(9)12/h1-6H
<b>InchiKey:</b>	GTILXPRQNNYDHT-UHFFFAOYSA-N
<b>Formula:</b>	C10H6Br2
<b>SMILES:</b>	BrC1cc2ccccc2cc1Br
<b>Mol. weight [g/mol]:</b>	285.96
<b>CAS:</b>	13214-70-5

## Physical Properties

Property code	Value	Unit	Source
gf	261.76	kJ/mol	Joback Method
hf	207.59	kJ/mol	Joback Method
hfus	22.51	kJ/mol	Joback Method
hvap	55.96	kJ/mol	Joback Method
log10ws	-5.60		Crippen Method
logp	4.365		Crippen Method
mcvol	143.540	ml/mol	McGowan Method
pc	4480.21	kPa	Joback Method
tb	616.14	K	Joback Method
tc	886.97	K	Joback Method
tf	406.22	K	Joback Method
vc	0.533	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	273.88	J/mol×K	616.14	Joback Method
cpg	283.96	J/mol×K	661.28	Joback Method
cpg	293.09	J/mol×K	706.42	Joback Method
cpg	301.39	J/mol×K	751.56	Joback Method
cpg	308.99	J/mol×K	796.69	Joback Method
cpg	316.02	J/mol×K	841.83	Joback Method
cpg	322.60	J/mol×K	886.97	Joback Method
dvisc	0.0013543	Paxs	406.22	Joback Method

dvisc	0.0010195	Paxs	441.21	Joback Method
dvisc	0.0008001	Paxs	476.19	Joback Method
dvisc	0.0006491	Paxs	511.18	Joback Method
dvisc	0.0005409	Paxs	546.17	Joback Method
dvisc	0.0004608	Paxs	581.15	Joback Method
dvisc	0.0003997	Paxs	616.14	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13214705&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13214705&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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