

# 2,6-Dihydroxynaphthalene

<b>Other names:</b>	2,6-Naphthalenediol C.I. 76640 Naphthalene-2,6-diol 2-Hydroxy-6-naphthol 2,6-Naphthohydroquinone
<b>Inchi:</b>	InChI=1S/C10H8O2/c11-9-3-1-7-5-10(12)4-2-8(7)6-9/h1-6,11-12H
<b>InchiKey:</b>	MNZMMCXVIXORAQL-UHFFFAOYSA-N
<b>Formula:</b>	C10H8O2
<b>SMILES:</b>	Oc1ccc2cc(O)ccc2c1
<b>Mol. weight [g/mol]:</b>	160.17
<b>CAS:</b>	581-43-1

## Physical Properties

Property code	Value	Unit	Source
gf	-56.86	kJ/mol	Joback Method
hf	-176.75	kJ/mol	Joback Method
hfus	24.28	kJ/mol	Joback Method
hvap	67.80	kJ/mol	Joback Method
log10ws	-2.37		Crippen Method
logp	2.251		Crippen Method
mccvol	120.280	ml/mol	McGowan Method
pc	5880.91	kPa	Joback Method
tb	635.10	K	Joback Method
tc	895.57	K	Joback Method
tf	485.02	K	Joback Method
vc	0.342	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	299.04	J/mol×K	635.10	Joback Method
cpg	308.81	J/mol×K	678.51	Joback Method
cpg	317.75	J/mol×K	721.92	Joback Method
cpg	326.08	J/mol×K	765.34	Joback Method

cpg	334.05	J/mol×K	808.75	Joback Method
cpg	341.88	J/mol×K	852.16	Joback Method
cpg	349.82	J/mol×K	895.57	Joback Method
dvisc	0.0001338	Paxs	485.02	Joback Method
dvisc	0.0000728	Paxs	510.03	Joback Method
dvisc	0.0000420	Paxs	535.05	Joback Method
dvisc	0.0000254	Paxs	560.06	Joback Method
dvisc	0.0000161	Paxs	585.07	Joback Method
dvisc	0.0000105	Paxs	610.09	Joback Method
dvisc	0.0000071	Paxs	635.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C581431&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C581431&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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