



dvisc	0.0000011	Paxs	1068.52	Joback Method
dvisc	0.0000014	Paxs	996.96	Joback Method
dvisc	0.0000021	Paxs	925.41	Joback Method
dvisc	0.0000031	Paxs	853.85	Joback Method
dvisc	0.0000051	Paxs	782.30	Joback Method
dvisc	0.0000091	Paxs	710.74	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R578113&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R578113&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/77-256-3/Butyl-2-5-8-11-14-17-20-23-26-29-32-undecaoxatetatriacontan-34-oate.pdf>

Generated by Cheméo on 2024-04-28 20:06:12.322008666 +0000 UTC m=+16624021.242585978.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.