

# Hexacosane, 4-methyl

<b>Other names:</b>	4-methylhexacosane
<b>Inchi:</b>	InChI=1S/C27H56/c1-4-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23-24-26-27(3
<b>InchiKey:</b>	RWVONPYEUYBBJH-UHFFFAOYSA-N
<b>Formula:</b>	C27H56
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCC(C)CCC
<b>Mol. weight [g/mol]:</b>	380.73

## Physical Properties

Property code	Value	Unit	Source
gf	174.02	kJ/mol	Joback Method
hf	-605.89	kJ/mol	Joback Method
hfus	62.16	kJ/mol	Joback Method
hvap	75.31	kJ/mol	Joback Method
log10ws	-10.88		Crippen Method
logp	10.635		Crippen Method
mcvol	391.290	ml/mol	McGowan Method
pc	691.79	kPa	Joback Method
rinpol	2662.00		NIST Webbook
rinpol	2660.00		NIST Webbook
rinpol	2658.00		NIST Webbook
rinpol	2660.70		NIST Webbook
rinpol	2662.00		NIST Webbook
rinpol	2655.00		NIST Webbook
rinpol	2663.00		NIST Webbook
rinpol	2677.00		NIST Webbook
rinpol	2658.00		NIST Webbook
ripol	2654.00		NIST Webbook
ripol	2654.00		NIST Webbook
tb	816.72	K	Joback Method
tc	1000.04	K	Joback Method
tf	379.05	K	Joback Method
vc	1.542	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1282.19	J/molxK	816.72	Joback Method
cpg	1306.88	J/molxK	847.27	Joback Method
cpg	1330.33	J/molxK	877.83	Joback Method
cpg	1352.60	J/molxK	908.38	Joback Method
cpg	1373.74	J/molxK	938.93	Joback Method
cpg	1393.80	J/molxK	969.49	Joback Method
cpg	1412.85	J/molxK	1000.04	Joback Method
dvisc	0.0020683	Paxs	379.05	Joback Method
dvisc	0.0006041	Paxs	451.99	Joback Method
dvisc	0.0002484	Paxs	524.94	Joback Method
dvisc	0.0001269	Paxs	597.88	Joback Method
dvisc	0.0000750	Paxs	670.83	Joback Method
dvisc	0.0000492	Paxs	743.77	Joback Method
dvisc	0.0000347	Paxs	816.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R248550&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R248550&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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