

# Acetophenone, 3'-allyl-4'-hydroxy-

<b>Other names:</b>	Ethanone, 1-(4-hydroxy-3-(2-propenyl)phenyl)- 3'-Allyl-4'-hydroxyacetophenone 3-Allyl-4-hydroxy-acetophenon 3-Allyl-4-hydroxy-acetophenone
<b>Inchi:</b>	InChI=1S/C11H12O2/c1-3-4-10-7-9(8(2)12)5-6-11(10)13/h3,5-7,13H,1,4H2,2H3
<b>InchiKey:</b>	XVTCWUFLNLZPEJ-UHFFFAOYSA-N
<b>Formula:</b>	C11H12O2
<b>SMILES:</b>	<chem>C=CCc1cc(C(C)=O)ccc1O</chem>
<b>Mol. weight [g/mol]:</b>	176.21
<b>CAS:</b>	1132-05-4

## Physical Properties

Property code	Value	Unit	Source
gf	-51.18	kJ/mol	Joback Method
hf	-209.77	kJ/mol	Joback Method
hfus	24.00	kJ/mol	Joback Method
hvap	62.11	kJ/mol	Joback Method
log10ws	-2.75		Crippen Method
logp	2.323		Crippen Method
mcvol	145.230	ml/mol	McGowan Method
pc	3443.98	kPa	Joback Method
tb	613.91	K	Joback Method
tc	841.77	K	Joback Method
tf	412.56	K	Joback Method
vc	0.496	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	352.40	J/molxK	613.91	Joback Method
cpg	364.55	J/molxK	651.89	Joback Method
cpg	375.87	J/molxK	689.86	Joback Method
cpg	386.45	J/molxK	727.84	Joback Method
cpg	396.37	J/molxK	765.82	Joback Method

cpg	405.71	J/molxK	803.79	Joback Method
cpg	414.58	J/molxK	841.77	Joback Method
dvisc	0.0008992	Paxs	412.56	Joback Method
dvisc	0.0004319	Paxs	446.12	Joback Method
dvisc	0.0002298	Paxs	479.68	Joback Method
dvisc	0.0001328	Paxs	513.24	Joback Method
dvisc	0.0000821	Paxs	546.79	Joback Method
dvisc	0.0000537	Paxs	580.35	Joback Method
dvisc	0.0000367	Paxs	613.91	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1132054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1132054&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/77-306-7/Acetophenone-3-allyl-4-hydroxy.pdf>

Generated by Cheméo on 2024-08-10 10:13:23.49636691 +0000 UTC m=+1974672.743472255.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.