

# 8b,16b-Dihydrodibenzo[g,p]chrysene mixture of cis and trans

**Inchi:** InChI=1S/C26H18/c1-5-13-21-17(9-1)18-10-2-6-14-22(18)26-24-16-8-4-12-20(24)19-11-3  
**InchiKey:** BUBKFBKPMZRIJ-UHFFFAOYSA-N  
**Formula:** C26H18  
**SMILES:** c1ccc2c(c1)-c1cccc1C1c3cccc3-c3cccc3C21  
**Mol. weight [g/mol]:** 330.42  
**CAS:** 33741-19-4

## Physical Properties

Property code	Value	Unit	Source
chs	-13178.00	kJ/mol	NIST Webbook
gf	749.06	kJ/mol	Joback Method
hf	490.51	kJ/mol	Joback Method
hfus	42.37	kJ/mol	Joback Method
hsub	182.00	kJ/mol	NIST Webbook
hvap	84.36	kJ/mol	Joback Method
log10ws	-8.95		Crippen Method
logp	6.611		Crippen Method
mcvol	260.440	ml/mol	McGowan Method
pc	1906.90	kPa	Joback Method
tb	917.32	K	Joback Method
tc	1188.78	K	Joback Method
tf	588.50	K	Joback Method
vc	1.006	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	795.71	J/molxK	917.32	Joback Method
cpg	812.87	J/molxK	962.56	Joback Method
cpg	829.68	J/molxK	1007.81	Joback Method
cpg	846.48	J/molxK	1053.05	Joback Method
cpg	863.61	J/molxK	1098.29	Joback Method
cpg	881.43	J/molxK	1143.53	Joback Method
cpg	900.26	J/molxK	1188.78	Joback Method

dvisc	0.0042000	Paxs	588.50	Joback Method
dvisc	0.0037395	Paxs	643.30	Joback Method
dvisc	0.0033909	Paxs	698.11	Joback Method
dvisc	0.0031188	Paxs	752.91	Joback Method
dvisc	0.0029013	Paxs	807.71	Joback Method
dvisc	0.0027239	Paxs	862.52	Joback Method
dvisc	0.0025767	Paxs	917.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33741194&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33741194&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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