

# Heptapropylene glycol, monoallyl ether, acetate

**Inchi:** InChI=1S/C26H50O9/c1-10-11-28-12-19(2)29-13-20(3)30-14-21(4)31-15-22(5)32-16-23(6)24-25(7)26  
**InchiKey:** ADBBXUXKDHMXJJ-UHFFFAOYSA-N  
**Formula:** C26H50O9  
**SMILES:** C=CCOCC(C)OCC(C)OCC(C)OCC(C)OCC(C)OCC(C)OCC(C)OC(C)=O  
**Mol. weight [g/mol]:** 506.67

## Physical Properties

Property code	Value	Unit	Source
gf	-730.12	kJ/mol	Joback Method
hf	-1661.84	kJ/mol	Joback Method
hfus	48.26	kJ/mol	Joback Method
hvap	96.11	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.571		Crippen Method
mcvol	421.430	ml/mol	McGowan Method
pc	734.82	kPa	Joback Method
rinpol	2653.00		NIST Webbook
rinpol	2656.00		NIST Webbook
rinpol	2655.00		NIST Webbook
rinpol	2658.00		NIST Webbook
rinpol	2657.00		NIST Webbook
rinpol	2660.00		NIST Webbook
rinpol	2660.00		NIST Webbook
rinpol	2657.00		NIST Webbook
rinpol	2653.00		NIST Webbook
rinpol	2654.00		NIST Webbook
rinpol	2658.00		NIST Webbook
rinpol	2660.00		NIST Webbook
rinpol	2656.00		NIST Webbook
rinpol	2656.00		NIST Webbook
tb	1021.11	K	Joback Method
tc	1263.48	K	Joback Method
tf	503.79	K	Joback Method
vc	1.581	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1483.05	J/molxK	1021.11	Joback Method
cpg	1499.98	J/molxK	1061.50	Joback Method
cpg	1513.78	J/molxK	1101.90	Joback Method
cpg	1524.40	J/molxK	1142.29	Joback Method
cpg	1531.77	J/molxK	1182.69	Joback Method
cpg	1535.85	J/molxK	1223.08	Joback Method
cpg	1536.58	J/molxK	1263.48	Joback Method
dvisc	0.0001325	Paxs	503.79	Joback Method
dvisc	0.0000394	Paxs	590.01	Joback Method
dvisc	0.0000160	Paxs	676.23	Joback Method
dvisc	0.0000079	Paxs	762.45	Joback Method
dvisc	0.0000045	Paxs	848.67	Joback Method
dvisc	0.0000029	Paxs	934.89	Joback Method
dvisc	0.0000020	Paxs	1021.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R152102&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R152102&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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