

# Fumaric acid, 2,6-dimethoxyphenyl dodec-2-en-1-yl ester

<b>Inchi:</b>	InChI=1S/C24H34O6/c1-4-5-6-7-8-9-10-11-12-13-19-29-22(25)17-18-23(26)30-24-20(27)
<b>InchiKey:</b>	FHYHDLDYNRJLHP-HUXYGNSESA-N
<b>Formula:</b>	C24H34O6
<b>SMILES:</b>	CCCCCCCCC=CCOC(=O)C=CC(=O)Oc1c(OC)cccc1OC
<b>Mol. weight [g/mol]:</b>	418.52

## Physical Properties

Property code	Value	Unit	Source
gf	-273.05	kJ/mol	Joback Method
hf	-844.70	kJ/mol	Joback Method
hfus	59.53	kJ/mol	Joback Method
hvap	95.67	kJ/mol	Joback Method
log10ws	-6.46		Crippen Method
logp	5.406		Crippen Method
mcvol	343.280	ml/mol	McGowan Method
pc	1068.66	kPa	Joback Method
rinpol	3054.00		NIST Webbook
tb	990.90	K	Joback Method
tc	1213.23	K	Joback Method
tf	590.32	K	Joback Method
vc	1.315	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1120.01	J/molxK	990.90	Joback Method
cpg	1179.73	J/molxK	1176.17	Joback Method
cpg	1170.58	J/molxK	1139.12	Joback Method
cpg	1160.07	J/molxK	1102.06	Joback Method
cpg	1148.16	J/molxK	1065.01	Joback Method
cpg	1134.81	J/molxK	1027.95	Joback Method
cpg	1187.54	J/molxK	1213.23	Joback Method
dvisc	0.0000133	Paxs	990.90	Joback Method
dvisc	0.0000171	Paxs	924.14	Joback Method

dvisc	0.0000228	Paxs	857.37	Joback Method
dvisc	0.0000320	Paxs	790.61	Joback Method
dvisc	0.0000477	Paxs	723.85	Joback Method
dvisc	0.0000772	Paxs	657.08	Joback Method
dvisc	0.0001395	Paxs	590.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405762&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405762&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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