

# Benzene, (1-octyldecyl)-

<b>Other names:</b>	Octadecane, 9-phenyl- 10-Phenyloctadecane 9-phenyloctadecane
<b>Inchi:</b>	InChI=1S/C24H42/c1-3-5-7-9-11-13-16-20-23(24-21-17-14-18-22-24)19-15-12-10-8-6-4-
<b>InchiKey:</b>	DPNNDICPUBVFIZ-UHFFFAOYSA-N
<b>Formula:</b>	C24H42
<b>SMILES:</b>	CCCCCCCCC(CCCCCC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	330.59
<b>CAS:</b>	10596-71-1

## Physical Properties

Property code	Value	Unit	Source
gf	261.17	kJ/mol	Joback Method
hf	-307.44	kJ/mol	Joback Method
hfus	48.43	kJ/mol	Joback Method
hvap	70.91	kJ/mol	Joback Method
log10ws	-8.94		Crippen Method
logp	8.662		Crippen Method
mcvol	325.260	ml/mol	McGowan Method
pc	987.02	kPa	Joback Method
tb	774.76	K	Joback Method
tc	960.07	K	Joback Method
tf	371.66	K	Joback Method
vc	1.266	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1007.54	J/mol×K	774.76	Joback Method
cpg	1028.90	J/mol×K	805.64	Joback Method
cpg	1049.13	J/mol×K	836.53	Joback Method
cpg	1068.28	J/mol×K	867.41	Joback Method
cpg	1086.41	J/mol×K	898.30	Joback Method
cpg	1103.57	J/mol×K	929.18	Joback Method

cpg	1119.81	J/mol×K	960.07	Joback Method
dvisc	0.0021815	Paxs	371.66	Joback Method
dvisc	0.0007349	Paxs	438.84	Joback Method
dvisc	0.0003305	Paxs	506.03	Joback Method
dvisc	0.0001792	Paxs	573.21	Joback Method
dvisc	0.0001105	Paxs	640.39	Joback Method
dvisc	0.0000747	Paxs	707.58	Joback Method
dvisc	0.0000540	Paxs	774.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10596711&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10596711&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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