

Isophthalic acid, di(pentafluorobenzyl) ester

Inchi: InChI=1S/C22H8F10O4/c23-11-9(12(24)16(28)19(31)15(11)27)5-35-21(33)7-2-1-3-8(4-7)
InchiKey: KCNITRSOMNWVTA-UHFFFAOYSA-N
Formula: C22H8F10O4
SMILES: O=C(OCc1c(F)c(F)c(F)c(F)c1F)c1cccc(C(=O)OCc2c(F)c(F)c(F)c(F)c2F)c1
Mol. weight [g/mol]: 526.28

Physical Properties

Property code	Value	Unit	Source
gf	-2050.28	kJ/mol	Joback Method
hf	-2364.69	kJ/mol	Joback Method
hfus	66.95	kJ/mol	Joback Method
hvap	88.82	kJ/mol	Joback Method
log10ws	-9.49		Crippen Method
logp	5.792		Crippen Method
mvol	282.140	ml/mol	McGowan Method
pc	1222.55	kPa	Joback Method
tb	982.86	K	Joback Method
tc	1203.41	K	Joback Method
tf	704.90	K	Joback Method
vc	1.171	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	847.22	J/mol×K	982.86	Joback Method
cpg	855.09	J/mol×K	1019.62	Joback Method
cpg	861.63	J/mol×K	1056.38	Joback Method
cpg	866.84	J/mol×K	1093.14	Joback Method
cpg	870.72	J/mol×K	1129.90	Joback Method
cpg	873.25	J/mol×K	1166.65	Joback Method
cpg	874.44	J/mol×K	1203.41	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U344513&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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