

Pentadecafluorooctanoic acid, tridec-2-yn-1-yl ester

Inchi:	InChI=1S/C21H23F15O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-38-14(37)15(22,23)16(24,25)1
InchiKey:	GVXWEJCI BRQRKO-UHFFFAOYSA-N
Formula:	C ₂₁ H ₂₃ F ₁₅ O ₂
SMILES:	CCCCCCCCCCC#CCOC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	592.38

Physical Properties

Property code	Value	Unit	Source
gf	-2807.45	kJ/mol	Joback Method
hf	-3452.17	kJ/mol	Joback Method
hfus	50.36	kJ/mol	Joback Method
hvap	52.32	kJ/mol	Joback Method
log10ws	-9.82		Crippen Method
logp	8.438		Crippen Method
mcvol	332.140	ml/mol	McGowan Method
pc	784.63	kPa	Joback Method
rinpol	1738.00		NIST Webbook
tb	731.61	K	Joback Method
tc	895.75	K	Joback Method
tf	530.48	K	Joback Method
vc	1.391	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1048.93	J/mol×K	731.61	Joback Method
cpg	1064.10	J/mol×K	758.97	Joback Method
cpg	1078.29	J/mol×K	786.32	Joback Method
cpg	1091.58	J/mol×K	813.68	Joback Method
cpg	1104.07	J/mol×K	841.04	Joback Method
cpg	1115.85	J/mol×K	868.40	Joback Method
cpg	1127.01	J/mol×K	895.75	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406814&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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