

# (3R,4R)-3-(Benzo[d][1,3]dioxol-5-ylmethyl)-4-((7-m

<b>Inchi:</b>	InChI=1S/C21H20O7/c1-23-18-7-13(8-19-20(18)28-11-27-19)4-14-9-24-21(22)15(14)5-1
<b>InchiKey:</b>	GBQHDHMFTXPYAM-CABCVRRESA-N
<b>Formula:</b>	C21H20O7
<b>SMILES:</b>	<chem>COc1cc(CC2COC(=O)C2Cc2ccc3c(c2)OCO3)cc2c1OCO2</chem>
<b>Mol. weight [g/mol]:</b>	384.38
<b>CAS:</b>	944284-56-4

## Physical Properties

Property code	Value	Unit	Source
gf	-189.82	kJ/mol	Joback Method
hf	-764.56	kJ/mol	Joback Method
hfus	66.01	kJ/mol	Joback Method
hvap	99.80	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	2.727		Crippen Method
mvol	263.440	ml/mol	McGowan Method
pc	2064.24	kPa	Joback Method
rinpol	3265.40		NIST Webbook
tb	1016.56	K	Joback Method
tc	1275.56	K	Joback Method
tf	716.19	K	Joback Method
vc	0.982	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	907.17	J/mol×K	1016.56	Joback Method
cpg	920.14	J/mol×K	1059.73	Joback Method
cpg	931.82	J/mol×K	1102.89	Joback Method
cpg	942.33	J/mol×K	1146.06	Joback Method
cpg	951.76	J/mol×K	1189.23	Joback Method
cpg	960.23	J/mol×K	1232.39	Joback Method
cpg	967.84	J/mol×K	1275.56	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C944284564&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C944284564&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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