

# 1,1'-Biphenyl, 3,4',5-triisopropyl

<b>Inchi:</b>	InChI=1S/C21H28/c1-14(2)17-7-9-18(10-8-17)21-12-19(15(3)4)11-20(13-21)16(5)6/h7-16
<b>InchiKey:</b>	QNOLWWWIKARVDL-UHFFFAOYSA-N
<b>Formula:</b>	C21H28
<b>SMILES:</b>	CC(C)c1ccc(-c2cc(C(C)C)cc(C(C)C)c2)cc1
<b>Mol. weight [g/mol]:</b>	280.45

## Physical Properties

Property code	Value	Unit	Source
gf	314.55	kJ/mol	Joback Method
hf	-53.96	kJ/mol	Joback Method
hfus	26.49	kJ/mol	Joback Method
hvap	67.71	kJ/mol	Joback Method
log10ws	-7.63		Crippen Method
logp	6.724		Crippen Method
mvol	259.230	ml/mol	McGowan Method
pc	1487.29	kPa	Joback Method
rinpol	2069.00		NIST Webbook
rinpol	2069.00		NIST Webbook
tb	746.86	K	Joback Method
tc	971.85	K	Joback Method
tf	371.83	K	Joback Method
vc	0.978	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	745.82	J/molxK	746.86	Joback Method
cpg	766.09	J/molxK	784.36	Joback Method
cpg	785.00	J/molxK	821.86	Joback Method
cpg	802.60	J/molxK	859.35	Joback Method
cpg	818.97	J/molxK	896.85	Joback Method
cpg	834.17	J/molxK	934.35	Joback Method
cpg	848.27	J/molxK	971.85	Joback Method
dvisc	0.0015830	Paxs	371.83	Joback Method

dvisc	0.0006464	Paxs	434.34	Joback Method
dvisc	0.0003306	Paxs	496.84	Joback Method
dvisc	0.0001965	Paxs	559.35	Joback Method
dvisc	0.0001296	Paxs	621.85	Joback Method
dvisc	0.0000923	Paxs	684.36	Joback Method
dvisc	0.0000695	Paxs	746.86	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R146291&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R146291&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/77-494-9/1-1-Biphenyl-3-4-5-triisopropyl.pdf>

Generated by Cheméo on 2024-04-29 00:05:20.344912966 +0000 UTC m=+16638369.265490281.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.