

# Benzenemethanol, «alpha»-phenyl-«alpha»-(2-phenylethynyl)-

Other names:

1,1,3-triphenylpropargyl alcohol

Diphenyl(phenylethynyl)carbinol

Inchi: InChI=1S/C21H16O/c22-21(19-12-6-2-7-13-19,20-14-8-3-9-15-20)17-16-18-10-4-1-5-11-

InchiKey: VWRQCJRTHKUVNF-UHFFFAOYSA-N

Formula: C21H16O

SMILES: OC(C#Cc1ccccc1)(c1ccccc1)c1ccccc1

Mol. weight [g/mol]: 284.35

CAS: 1522-13-0

## Physical Properties

Property code	Value	Unit	Source
chs	-10720.00	kJ/mol	NIST Webbook
gf	531.99	kJ/mol	Joback Method
hf	344.14	kJ/mol	Joback Method
hfs	194.30	kJ/mol	NIST Webbook
hfus	32.06	kJ/mol	Joback Method
hvap	86.70	kJ/mol	Joback Method
log10ws	-5.35		Crippen Method
logp	3.974		Crippen Method
mcvol	232.740	ml/mol	McGowan Method
pc	2576.72	kPa	Joback Method
tb	857.87	K	Joback Method
tc	1124.23	K	Joback Method
tf	575.03	K	Joback Method
vc	0.858	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	664.89	J/molxK	857.87	Joback Method
cpg	679.03	J/molxK	902.26	Joback Method
cpg	691.95	J/molxK	946.66	Joback Method
cpg	703.88	J/molxK	991.05	Joback Method
cpg	715.00	J/molxK	1035.44	Joback Method

cpg	725.51	J/mol×K	1079.83	Joback Method
cpg	735.60	J/mol×K	1124.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1522130&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1522130&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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