

# 2-Eicosanone

Inchi:

InChI=1S/C20H40O/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20(2)21/h3-19H2,1

InchiKey:

GARHHHWGJIXLDK-UHFFFAOYSA-N

Formula:

C20H40O

SMILES:

CCCCCCCCCCCCCCCCCCC(C)=O

Mol. weight [g/mol]:

296.53

CAS:

29703-52-4

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -11.40  | kJ/mol  | Joback Method  |
| hf            | -568.71 | kJ/mol  | Joback Method  |
| hfus          | 49.16   | kJ/mol  | Joback Method  |
| hvap          | 66.86   | kJ/mol  | Joback Method  |
| log10ws       | -7.47   |         | Crippen Method |
| logp          | 7.227   |         | Crippen Method |
| mcvol         | 294.230 | ml/mol  | McGowan Method |
| pc            | 1057.57 | kPa     | Joback Method  |
| rinpol        | 2206.00 |         | NIST Webbook   |
| rinpol        | 362.51  |         | NIST Webbook   |
| rinpol        | 2206.00 |         | NIST Webbook   |
| tb            | 710.87  | K       | Joback Method  |
| tc            | 880.03  | K       | Joback Method  |
| tf            | 365.09  | K       | Joback Method  |
| vc            | 1.161   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 875.44 | J/mol×K | 710.87          | Joback Method |
| cpg           | 895.27 | J/mol×K | 739.06          | Joback Method |
| cpg           | 914.21 | J/mol×K | 767.26          | Joback Method |
| cpg           | 932.27 | J/mol×K | 795.45          | Joback Method |
| cpg           | 949.51 | J/mol×K | 823.64          | Joback Method |
| cpg           | 965.93 | J/mol×K | 851.83          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 981.58    | J/mol×K | 880.03 | Joback Method |
| dvisc | 0.0024837 | Pa×s    | 365.09 | Joback Method |
| dvisc | 0.0009987 | Pa×s    | 422.72 | Joback Method |
| dvisc | 0.0004997 | Pa×s    | 480.35 | Joback Method |
| dvisc | 0.0002900 | Pa×s    | 537.98 | Joback Method |
| dvisc | 0.0001870 | Pa×s    | 595.61 | Joback Method |
| dvisc | 0.0001303 | Pa×s    | 653.24 | Joback Method |
| dvisc | 0.0000962 | Pa×s    | 710.87 | Joback Method |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.47541e+01                   |
| Coeff. B                    | -5.27580e+03                  |
| Coeff. C                    | -1.15748e+02                  |
| Temperature range (K), min. | 480.44                        |
| Temperature range (K), max. | 674.47                        |

## Sources

|                                      |                                                                                                                                                                                         |
|--------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Crippen Method:                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                                                               |
| Crippen Method:                      | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                                                                       |
| Joback Method:                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                                                                   |
| McGowan Method:                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                                                                   |
| NIST Webbook:                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C29703524&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C29703524&amp;Units=SI</a>                                           |
| The Yaws Handbook of Vapor Pressure: | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |

## Legend

|        |                                              |
|--------|----------------------------------------------|
| cpg:   | Ideal gas heat capacity                      |
| dvisc: | Dynamic viscosity                            |
| gf:    | Standard Gibbs free energy of formation      |
| hf:    | Enthalpy of formation at standard conditions |

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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