

# Eicosane, 1-iodo-

<b>Other names:</b>	1-Iodoeicosane
<b>Inchi:</b>	InChI=1S/C20H41I/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21/h2-20H2,1
<b>InchiKey:</b>	UYBWORFIJZQKEP-UHFFFAOYSA-N
<b>Formula:</b>	C20H41I
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	408.44
<b>CAS:</b>	34994-81-5

## Physical Properties

Property code	Value	Unit	Source
gf	175.64	kJ/mol	Joback Method
hf	-379.26	kJ/mol	Joback Method
hfus	51.96	kJ/mol	Joback Method
hvap	118.50	kJ/mol	NIST Webbook
log10ws	-9.14		Crippen Method
logp	8.463		Crippen Method
mcvol	318.480	ml/mol	McGowan Method
pc	1001.44	kPa	Joback Method
rinpol	2519.00		NIST Webbook
rinpol	2519.00		NIST Webbook
tb	750.14	K	Joback Method
tc	930.35	K	Joback Method
tf	314.60 ± 1.50	K	NIST Webbook
vc	1.244	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1027.09	J/mol×K	930.35	Joback Method
cpg	1011.59	J/mol×K	900.31	Joback Method
cpg	995.33	J/mol×K	870.28	Joback Method
cpg	978.25	J/mol×K	840.24	Joback Method
cpg	960.32	J/mol×K	810.21	Joback Method
cpg	941.49	J/mol×K	780.17	Joback Method

cpg	921.72	J/mol×K	750.14	Joback Method
dvisc	0.0022797	Paxs	373.22	Joback Method
dvisc	0.0000737	Paxs	750.14	Joback Method
dvisc	0.0001005	Paxs	687.32	Joback Method
dvisc	0.0001460	Paxs	624.50	Joback Method
dvisc	0.0002305	Paxs	561.68	Joback Method
dvisc	0.0004082	Paxs	498.86	Joback Method
dvisc	0.0008522	Paxs	436.04	Joback Method
hvapt	80.90	kJ/mol	594.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.62087e+01
Coeff. B	-6.19153e+03
Coeff. C	-1.30232e+02
Temperature range (K), min.	519.12
Temperature range (K), max.	698.41

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C34994815&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C34994815&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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