

# 2,2'-Binaphthalene, 1,1',2,2',3,3',4,4'-octahydro-

<b>Other names:</b>	2,2'-Binaphthyl, 1,1',2,2',3,3',4,4'-octahydro-2,2'-Bitetralin
<b>Inchi:</b>	InChI=1S/C20H22/c1-3-7-17-13-19(11-9-15(17)5-1)20-12-10-16-6-2-4-8-18(16)14-20/h1-
<b>InchiKey:</b>	BXELUHKITUQVQG-UHFFFAOYSA-N
<b>Formula:</b>	C20H22
<b>SMILES:</b>	<chem>c1ccc2c(c1)CCC(C1CCc3ccccc3C1)C2</chem>
<b>Mol. weight [g/mol]:</b>	262.39
<b>CAS:</b>	27426-98-8

## Physical Properties

Property code	Value	Unit	Source
gf	420.38	kJ/mol	Joback Method
hf	127.27	kJ/mol	Joback Method
hfus	26.93	kJ/mol	Joback Method
hvap	66.16	kJ/mol	Joback Method
log10ws	-5.64		Crippen Method
logp	4.597		Crippen Method
mcvol	223.420	ml/mol	McGowan Method
pc	2073.65	kPa	Joback Method
tb	742.34	K	Joback Method
tc	1001.79	K	Joback Method
tf	421.88	K	Joback Method
vc	0.838	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	663.89	J/molxK	742.34	Joback Method
cpg	771.64	J/molxK	1001.79	Joback Method
cpg	756.94	J/molxK	958.55	Joback Method
cpg	741.18	J/molxK	915.30	Joback Method
cpg	724.17	J/molxK	872.06	Joback Method
cpg	705.74	J/molxK	828.82	Joback Method
cpg	685.71	J/molxK	785.58	Joback Method

cpl	379.90	J/mol×K	298.15	NIST Webbook
dvisc	0.0004122	Paxs	742.34	Joback Method
dvisc	0.0004847	Paxs	688.93	Joback Method
dvisc	0.0005859	Paxs	635.52	Joback Method
dvisc	0.0007331	Paxs	582.11	Joback Method
dvisc	0.0009599	Paxs	528.70	Joback Method
dvisc	0.0013353	Paxs	475.29	Joback Method
dvisc	0.0020194	Paxs	421.88	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27426988&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27426988&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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